

MASTER FORMULA SHEET

CHAPTER 10:

PV = nRT (IDEAL GAS)

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2} \text{ (COMBINED GAS LAW)}$$

$$\lambda = \frac{c}{\nu} \text{ (WAVELENGTH)} \quad \text{OR} \quad \lambda = \frac{h}{mv} \text{ (DE BROGLIE WAVELENGTH)}$$

$$P_1 = P_2 = P_3 = \dots \text{ (TOTAL PRESSURE IN PARTIAL PRESSURES)}$$

$$P_i = \left(\frac{n_i}{n}\right) P_{\text{TOTAL}} \text{ (RELATIVE T. PARTIAL PRESSURE)}$$

$$M_{\text{RMS}} = \sqrt{\frac{3RT}{M}} \text{ (RMS SPEED OF MOLECULES)}$$

$$\frac{r_1}{r_2} = \sqrt{\frac{M_2}{M_1}} \text{ (RATE OF EFFUSION OF GASES)}$$

CHAPTER 11: IN INTERMOLECULAR FORCES

LPE < DIPOLE-DIPOLE < H-BONDING < VAN DER WAALS

IF WEIGHT OF A MOLECULE IS MUCH > ANOTHER, DISPERSION FORCES

$$\ln P = \frac{-\Delta H_{\text{vap}}}{RT} + C \text{ (CLAUZIUS-CLAPRON)}$$

WITH GRAPH: SLOPE = $-\Delta H_{\text{vap}}/R$

CHAPTER 12:

REC. CAT. / FORM. UNIT = AN. COORD. N

REC. CAT. / FORM. UNIT = AN. COORD. N

$$\frac{4\pi r^3}{3} \text{ (VOL. OF SPHERE)}$$

PACK. CUBE (1 ATOM) / 2 ATOMS (4 ATOMS) / 4 ATOMS (8 ATOMS)

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CHAPTER 15:

$$K_c = \frac{[D]^d [E]^e}{[A]^a [B]^b} \text{ (EQUILIBRIUM CONSTANT)}$$

(USES ONLY FOR CONCENT.)

$$K_p = \frac{(P_D)^d (P_E)^e}{(P_A)^a (P_B)^b} \text{ (EQUILIBRIUM CONSTANT)}$$

RELATING THE TWO EQS

$$K_p = K_c (RT)^{\Delta n} \text{ (RELATING THE TWO EQS)}$$

$$Q_c = \frac{[D]^d [E]^e}{[A]^a [B]^b} \text{ (REACTION QUOTIENT)}$$

IF $Q < K$, $Q > K$

IF $Q = K$, $Q > K$

IF $Q < K$, $Q > K$

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IF $Q < K$, $Q > K$

CHAPTER 17:

$$\Delta S = q_{\text{rev}}/T \text{ (FOR REVERSIBLE PROCESSES)}$$

2ND LAW OF THERMODYNAMICS:

$$\text{REV: } \Delta S_{\text{UNIV}} = \Delta S_{\text{SYS}} + \Delta S_{\text{SURR}} = 0$$

$$\text{REV: } \Delta S_{\text{UNIV}} = \Delta S_{\text{SYS}} + \Delta S_{\text{SURR}} > 0$$

$$\text{IRREV: } \Delta S_{\text{UNIV}} = \Delta S_{\text{SYS}} + \Delta S_{\text{SURR}} > 0$$

$$\Delta S = \int \frac{dq_{\text{rev}}}{T} \text{ (RELATING } \Delta S \text{ TO } q_{\text{rev}} \text{)}$$

$$\Delta S_{\text{SURR}} = \frac{-\Delta H_{\text{SYS}}}{T} \text{ (RELATING } \Delta S_{\text{SURR}} \text{ TO } \Delta H_{\text{SYS}} \text{)}$$

$$\Delta G = \Delta H - T\Delta S \text{ (GIBBS FREE ENERGY)}$$

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