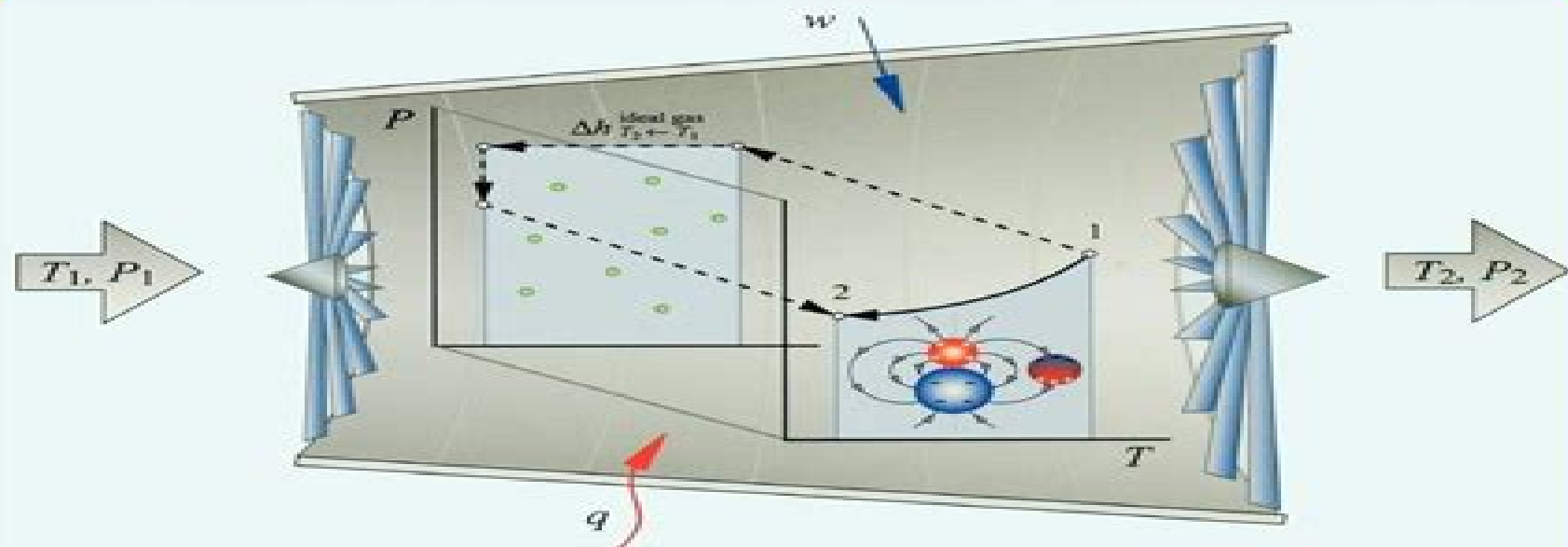




Engineering and Chemical Thermodynamics



Milo D. Koretsky

Koretsky Milo Engineering And Chemical Thermodynamics

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 and developing and improving technologies and products Concepts such as internal energy enthalpy entropy and Gibbs
 energy are explained using ideas and experiences familiar to students and realistic examples are given so the usefulness and
 pervasiveness of thermodynamics becomes apparent The worked examples illustrate key ideas and demonstrate important
 types of calculations and the problems at the end of chapters are designed to reinforce important concepts and show the
 broad range of applications Most can be solved using digitized data from open access databases and a spreadsheet Answers
 are provided for the numerical problems A particular theme of the book is the calculation of the equilibrium composition of
 systems both reactive and non reactive and this includes the principles of Gibbs energy minimization The overall approach
 leads to the intelligent use of thermodynamic software packages but while these are discussed and their use demonstrated
 they are not the focus of the book the aim being to provide the necessary foundations Another unique aspect is the inclusion
 of three applications chapters heat and energy aspects of processing the thermodynamics of metal production and recycling
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 science materials science and metallurgy though it will be also useful for students undertaking courses in geology and
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 example in batteries For complex systems the number of independent reactions their nature have to be determined

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