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CHAPTER 15 (FIN MAN); CHAPTER 1 (MAN) INTRODUCTION TO MANAGERIAL ACCOUNTING

DISCUSSION QUESTIONS

1. Financial accounting and managerial accounting are different in several ways. Financial accounting information is reported in statements that are useful to persons or groups outside of a company. These statements objectively report the results of operations for fixed periods of time and the financial condition of the business under generally accepted accounting principles. Managerial accounting information uses both subjective and objective information to meet the specific needs of management. This non-GAAP information can be reported periodically or as needed by management and can be reported for the entire entity or for segments of the organization. This information includes (i) historical data, which provide objective measures of past operations, and (ii) estimated data, which provide subjective estimates about future decisions.
2. **A.** A line department is directly involved in providing goods and services to customers, while a staff department provides service, assistance, or advice to line departments or other staff departments.
B. (1) Sales Department
(2) Personnel Department
3. Direct materials cost
4. Prime costs are the combination of direct materials and direct labor costs, while conversion costs are the combination of direct labor costs and factory overhead costs.
5. Product costs are composed of three elements of manufacturing costs: direct materials cost, direct labor cost, and factory overhead cost. These costs are treated as assets until the product is sold. Period costs consist of selling and administrative expenses that are used in generating revenue during the current period. They are recognized as expenses on the current period's income statement.
6. The three inventory accounts for a manufacturing business are as follows:
A. Finished goods inventory consists of completed (or finished) products that have not been sold.
B. Work in process inventory consists of the direct materials, direct labor, and factory overhead costs

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Mark Steyvers



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 $\text{kclO}_3 \rightarrow 2 \text{kcl} + 3 \text{o}_2$
 $\text{nacl} + \text{f}_2 \rightarrow \text{naf} + \text{cl}_2$
 $2 \text{h}_2 + \text{o}_2 \rightarrow 2 \text{h}_2\text{o}$
 $\text{pb} + \text{oh}^- + \text{hcl} \rightarrow \text{h}_2\text{o} + \text{pbcl}_2$
 $\text{albr}_3 + 3 \text{k}_2\text{so}_4 \rightarrow 6 \text{kbr} + \text{al}_2\text{so}_4 + 3 \text{ch}_4$
 $2 \text{o}_2 + \text{co}_2 + 2 \text{h}_2\text{o} + \text{c}_3\text{h}_8 \rightarrow 5 \text{o}_2 + 3 \text{co}_2 + 4 \text{h}_2\text{o} + \text{c}_8\text{h}_{18}$
 $25 \text{o}_2 + 16 \text{co}_2 + 18 \text{h}_2\text{o}$

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 solid calcium carbonate is heated and decomposes to solid calcium oxide and carbon dioxide gas
 gaseous butane C_4H_{10} reacts with

diatomic oxygen gas to yield gaseous carbon dioxide and water vapor

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web mar 14 2023 the key to balancing chemical equations the ultimate goal for balancing chemical equations is to make both sides of the reaction the reactants and the products equal in the number of atoms per element this stems from the universal law of the conservation of mass which states that matter can neither be created nor destroyed

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