

**Acid - Base Titration Lab**

<b>Knowledge</b>		<b>/16</b>			
<b>Introduction (Knowledge component)</b>		<b>/16</b>			
-acids and bases are well defined		0	1	2	3
-strong and weak acid/base is clearly explained		0	1	2	3
-neutralization is well explained		0	1	2	3
-good definition of titration and explanation		0	1	2	3
-relevant real-world application of acid-base chemistry described		0	1	2	3
-purpose of the lab		0	1		
<b>Communication</b>		<b>/12</b>			
<b>Title Page</b>		<b>/1</b>			
-Includes appropriate and descriptive title for the lab		0	1		
<b>Introduction (Communication component)</b>		<b>/4</b>			
-written in full paragraphs (no headings/questions)		0	1	2	
-facts are correctly cited		0	1	2	
<b>Results - Observation Tables</b>		<b>/5</b>			
- Appropriate titles at the top		0	1		
- Units are in the headings		0	1		
- Well organized and neat (ex. numbers are aligned)		0	1		
- Contains all relevant data and observations (including averages)		0	1	2	
<b>Overall clarity, neatness, formatting, spelling and grammar</b>		<b>/2</b>			
<b>Inquiry</b>		<b>/17</b>			
<b>Analysis</b>		<b>/4</b>			
-Balanced chemical reaction is complete		0	1		
-Full calculations showing how to determine the concentration of HCl		0	1	2	3
<b>Discussion</b>		<b>/9</b>			
-difference between end point and equivalence point		0	1	2	
-NaOH's hygroscopic nature and its effect on the results		0	1	2	3
-Two experimental sources of error discussed and explained		0	1	2	3 4
<b>Conclusion</b>		<b>/2</b>			
-Final statement summarizing results of the lab		0	1		
-One thing that was learned from the lab		0	1		
<b>References/Work Cited</b>		<b>/2</b>			
-Correctly written in APA format					

**Total : /45**

# K94749 Chemitry Mark Scheme

**Lingsheng Yao**



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