

A-Level Chemistry - Year 2

OCR Chemistry A - Paper 1 June - 2019

16 Sir Humphrey Davy discovered several elements including sodium, potassium, magnesium, calcium and strontium.

(a) Explain which block in the Periodic Table sodium and magnesium belong to.

(1)

(b) A sample of magnesium, $A_r = 24.305$, is found to consist of three isotopes. The accurate relative isotopic masses and % abundances of two of the isotopes are shown in the table.

Isotope	Relative isotopic mass	% abundance
^{24}Mg	23.985	78.99%
^{26}Mg	24.986	10.00%

Determine the relative isotopic mass of the third isotope of magnesium in the sample.

Give your answer to 5 significant figures.

relative isotopic mass = _____ (2)

Question - 16

OCR

International Centre for

Tuesday 4 June 2019 – Afternoon

A Level Chemistry A

H43201 Periodic table, elements and physical chemistry

Time allowed: 1 hour 15 minutes

We need from:

1. An answer sheet (OCR form 1) and a pencil.

We also need:

1. A ruler and a calculator.



Please write clearly in block capitals. Do not write in the brackets.

Centre number

Candidate number

First name(s)

Last name

INSTRUCTIONS

- 1. Use black ink. You may use an HB pencil for graphs and diagrams.
- 2. Answer all the questions.
- 3. Where appropriate, your answers should be supported with working. Marks may be given for a correct method even if the answer is incorrect.
- 4. Write your answer to each question in the space provided. If additional space is required, use the back page(s) at the end of this booklet. The question numbers must be clearly shown.

INFORMATION

- 1. The total mark for this paper is 100.
- 2. The marks for each question are shown in brackets ().
- 3. Quality of extended responses will be assessed in questions marked with an asterisk (*).
- 4. This document consists of 32 pages.

June 2013 Ocr Chemistry F321 Past Paper

Mike Smith



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Table of Contents June 2013 Ocr Chemistry F321 Past Paper

1. Understanding the eBook June 2013 Ocr Chemistry F321 Past Paper
 - The Rise of Digital Reading June 2013 Ocr Chemistry F321 Past Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying June 2013 Ocr Chemistry F321 Past Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an June 2013 Ocr Chemistry F321 Past Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from June 2013 Ocr Chemistry F321 Past Paper
 - Personalized Recommendations
 - June 2013 Ocr Chemistry F321 Past Paper User Reviews and Ratings
 - June 2013 Ocr Chemistry F321 Past Paper and Bestseller Lists
5. Accessing June 2013 Ocr Chemistry F321 Past Paper Free and Paid eBooks
 - June 2013 Ocr Chemistry F321 Past Paper Public Domain eBooks
 - June 2013 Ocr Chemistry F321 Past Paper eBook Subscription Services
 - June 2013 Ocr Chemistry F321 Past Paper Budget-Friendly Options
6. Navigating June 2013 Ocr Chemistry F321 Past Paper eBook Formats
 - ePub, PDF, MOBI, and More
 - June 2013 Ocr Chemistry F321 Past Paper Compatibility with Devices
 - June 2013 Ocr Chemistry F321 Past Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of June 2013 Ocr Chemistry F321 Past Paper
 - Highlighting and Note-Taking June 2013 Ocr Chemistry F321 Past Paper
 - Interactive Elements June 2013 Ocr Chemistry F321 Past Paper
8. Staying Engaged with June 2013 Ocr Chemistry F321 Past Paper

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers June 2013 Ocr Chemistry F321 Past Paper
- 9. Balancing eBooks and Physical Books June 2013 Ocr Chemistry F321 Past Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection June 2013 Ocr Chemistry F321 Past Paper
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine June 2013 Ocr Chemistry F321 Past Paper
 - Setting Reading Goals June 2013 Ocr Chemistry F321 Past Paper
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of June 2013 Ocr Chemistry F321 Past Paper
 - Fact-Checking eBook Content of June 2013 Ocr Chemistry F321 Past Paper
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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