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Tipuri de reacții în chimia anorganică

Manual pentru licee industriale cu profil de chimie, clasa a XI-a

N.O.	
+6	  $\text{Cr}_2\text{O}_7^{2-}$
+5	 
+4	
+3	  Cr_2O_3
+2	
+1	
0	
-1	
-2	
-3	

$(\text{NH}_4)_2\text{Cr}_2\text{O}_7$

N_2



NH_4^+

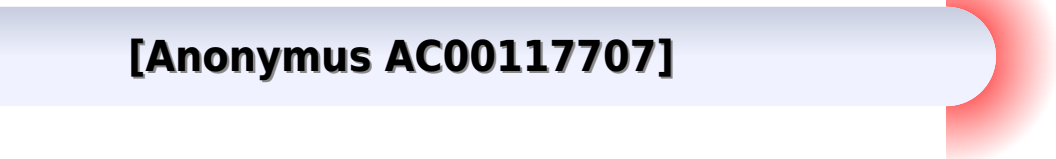


Diagram illustrating the oxidation states of Chromium (Cr) and Nitrogen (N) in various compounds. The central table shows the oxidation state (N.O.) on the left and the corresponding chemical formula on the right. Arrows indicate the change in oxidation state during the reaction of $(\text{NH}_4)_2\text{Cr}_2\text{O}_7$ to N_2 and NH_4^+ .

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[Anonymus AC00117707]



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Previously by Angelici this laboratory manual for an upper level undergraduate or graduate course in inorganic synthesis has for many years been the standard in the field In this newly revised third edition the manual has been extensively updated to reflect new developments in inorganic chemistry Twenty three experiments are divided into five sections solid state chemistry main group chemistry coordination chemistry organometallic chemistry and bioinorganic chemistry The included experiments are safe have been thoroughly tested to ensure reproducibility are illustrative of modern issues in inorganic chemistry and are capable of being performed in one or two laboratory periods of three or four hours Because facilities vary from school to school the authors have included a broad range of experiments to help provide a meaningful course in almost any academic setting Each clearly written illustrated experiment begins with an introduction that highlights the theme of the experiment often including a discussion of a particular characterization method that will be used followed by the experimental procedure a set of problems a listing of suggested Independent Studies and literature references

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