

Molarity Calculations

Name _____	Section _____	Date _____
Molarity Calculations		
Directions: Solve each problem for the indicated quantity.		
1. What is the molarity of 1.32 L of a solution of 0.3087 moles of KMnO_4 ?		
2. How many moles of NaOH are present in 2.36 L of a 1.871 M aqueous solution of NaOH ?		
3. What volume in Liters is an aqueous solution of HCl if it has a molarity of 0.850 M and contains 0.580 moles of HCl ?		
4. What is the molarity of 0.1000 moles of KI in 31.7 mL of aqueous solution?		
5. What is the volume in mL of 1.5000 moles of NaClO with a molarity of 0.6341 M?		
6. What is the molarity of an aqueous solution of 2.35 grams of NaCl in a solution with a volume of 1.914 L?		
7. What is the volume of an aqueous solution in mL if it has a molarity of 0.112 M and contains 49.08 grams of dissolved KBr ?		
8. How many grams of H_2CO_3 are dissolved in 2.32 L of an aqueous solution with a molarity of 0.7542 M?		
9. Find the volume of an aqueous solution of KNO_3 in L if it has a molarity of 1.89 M and contains 16.27 grams of KNO_3 . If the solution contains L of water, what is the volume of the dissolved KNO_3 ?		
10. Potassium sulfate has a density of 2.66 $\frac{\text{g}}{\text{mL}}$. What is the molarity of a solution of 7.337 grams of K_2SO_4 and 1.375 L of water?		

$$\text{Molarity} = \frac{\text{Moles of Solute}}{\text{Liters of Solution}}$$

$$M = \frac{\text{mol}}{L}$$

A CHEMISTRY WORKSHEET

10 PROBLEMS

WORKED ANSWER KEY INCLUDED

Molarity Worksheet Pogil Activities For Highschool Chemistry

Marco Cascella



Molarity Worksheet Pogil Activities For Highschool Chemistry:

POGIL Activities for High School Chemistry High School POGIL Initiative, 2012 Solving Molarity Problems, 2016

Extend your understanding of molarity by solving some typical problems encountered in the high school chemistry classroom

To foster your understanding of these problems you are asked to draw upon the quantitative reasoning skills you previously used *Teaching the Molarity Principle to High School Students* Wilfred Francis Borden, 1974 **Chemistry Lesson**

Plans, Study Guides, and Lecture Notes Doctor Chemistry, 2017-08-13 The Ultimate Guide to Learning or Teaching Chemistry This book contains the real lecture notes and slide of a highly effective high school and college Chemistry teacher Teachers Never plan another lesson again Students Ace your upcoming exam This series covers all of the topics of High School Chemistry and General Chemistry including Accuracy and Significant Figures Mixtures Metric System Bonding Atomic Theory Periodic Table VSEPR Ionic and Covalent Bonding Geometric Bonding The Mole and Molar Mass Equation Balancing Thermodynamics Stoichiometry States of Matter Gas Laws and Calculations Reaction Calculations Acids and Bases Limiting Reagents Redox and Electro Chemistry Organic Chemistry Basics

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In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Molarity Worksheet Pogil Activities For Highschool Chemistry**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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