

CHAPTER 7 REVIEW*Chemical Formulas and Chemical Compounds***SECTION 7-1****SHORT ANSWER** Answer the following questions in the space provided.

1. c In a Stock name such as iron(III) sulfate, the roman numeral tells us _____
(a) how many atoms of Fe are in one formula unit
(b) how many sulfate ions can be attached to the iron atom
(c) the charge on each Fe ion
(d) the total positive charge of the formula unit
2. c The result of changing a subscript in a correctly written chemical formula is to _____
(a) change the number of moles represented by the formula
(b) change the charges on the other ions in the compound
(c) change the formula so that it no longer represents the compound it previously represented
(d) have no effect on the formula
3. The explosive TNT has the molecular formula $C_7H_5(NO_2)_3$.
 4 elements a. How many elements make up this compound?
 6 oxygen atoms b. How many oxygen atoms are present in one molecule of $C_7H_5(NO_2)_3$?
 21 atoms c. How many atoms in total are present in one molecule of $C_7H_5(NO_2)_3$?
 4.2×10^{24} atoms d. How many atoms are present in a sample of 2×10^{23} molecules of $C_7H_5(NO_2)_3$?
4. How many atoms are present in each of these formula units?
 11 atoms a. $Ca(HCO_3)_2$
 45 atoms b. $C_{12}H_{22}O_{11}$
 10 atoms c. $Fe(ClO_2)_3$
 9 atoms d. $Fe(ClO_3)_2$
5. N_2O_5 a. What is the formula for the compound dinitrogen pentoxide?
 carbon(IV) sulfide b. What is the Stock name for the covalent compound CS_2 ?
 H_2SO_3 c. What is the formula for sulfurous acid?
 phosphoric acid d. What is the name for the acid H_3PO_4 ?

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**Society of Chemical Industry (Great
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Table of Contents Modern Chemistry Chapter Review Chemical Formulas And Compounds Answers

1. Understanding the eBook Modern Chemistry Chapter Review Chemical Formulas And Compounds Answers
 - The Rise of Digital Reading Modern Chemistry Chapter Review Chemical Formulas And Compounds Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Modern Chemistry Chapter Review Chemical Formulas And Compounds Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Modern Chemistry Chapter Review Chemical Formulas And Compounds Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Modern Chemistry Chapter Review Chemical Formulas And Compounds Answers
 - Personalized Recommendations

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- Modern Chemistry Chapter Review Chemical Formulas And Compounds Answers User Reviews and Ratings
- Modern Chemistry Chapter Review Chemical Formulas And Compounds Answers and Bestseller Lists
- 5. Accessing Modern Chemistry Chapter Review Chemical Formulas And Compounds Answers Free and Paid eBooks
 - Modern Chemistry Chapter Review Chemical Formulas And Compounds Answers Public Domain eBooks
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 - Modern Chemistry Chapter Review Chemical Formulas And Compounds Answers Budget-Friendly Options
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 - Modern Chemistry Chapter Review Chemical Formulas And Compounds Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Modern Chemistry Chapter Review Chemical Formulas And Compounds Answers
 - Highlighting and Note-Taking Modern Chemistry Chapter Review Chemical Formulas And Compounds Answers
 - Interactive Elements Modern Chemistry Chapter Review Chemical Formulas And Compounds Answers
- 8. Staying Engaged with Modern Chemistry Chapter Review Chemical Formulas And Compounds Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Modern Chemistry Chapter Review Chemical Formulas And Compounds Answers
- 9. Balancing eBooks and Physical Books Modern Chemistry Chapter Review Chemical Formulas And Compounds Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Modern Chemistry Chapter Review Chemical Formulas And Compounds Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Modern Chemistry Chapter Review Chemical Formulas And Compounds Answers
 - Setting Reading Goals Modern Chemistry Chapter Review Chemical Formulas And Compounds Answers

- Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Modern Chemistry Chapter Review Chemical Formulas And Compounds Answers
 - Fact-Checking eBook Content of Modern Chemistry Chapter Review Chemical Formulas And Compounds Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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