



11 How does the arrangement of particles in a solid change when it melts? What happens to the energy of the particles when it melts?

10 How does the arrangement of particles in a solid change when it melts? What happens to the energy of the particles when it melts?

9 What is the change of state when a solid melts?

8 What is the change of state when a solid melts?

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PARTICLE THEORY AND STATES OF MATTER

Particle theory explains the properties of matter and the changes of state. It is based on the idea that all matter is made up of tiny particles. These particles have different arrangements in different states of matter. The particles are constantly moving and colliding with each other. The arrangement of particles explains the properties of the matter.

Solid State: Particles are packed closely together. They have strong forces of attraction between them. They vibrate in fixed positions. They have low energy. They have a definite shape and volume. They are incompressible. They have a high melting point. They have a high boiling point. They have a high density. They have a high refractive index. They have a high thermal conductivity. They have a high electrical conductivity. They have a high magnetic permeability. They have a high optical transparency. They have a high acoustic transparency. They have a high thermal expansion coefficient. They have a high thermal contraction coefficient. They have a high thermal stability. They have a high thermal instability. They have a high thermal resistance. They have a high thermal conductance. They have a high thermal capacitance. They have a high thermal inductance. They have a high thermal capacitance. They have a high thermal inductance. They have a high thermal resistance. They have a high thermal conductance. They have a high thermal capacitance. They have a high thermal inductance.

Liquid State: Particles are packed less closely together. They have weaker forces of attraction between them. They move more freely. They have a definite volume. They have a definite shape. They have a definite density. They have a definite refractive index. They have a definite thermal conductivity. They have a definite electrical conductivity. They have a definite magnetic permeability. They have a definite optical transparency. They have a definite acoustic transparency. They have a definite thermal expansion coefficient. They have a definite thermal contraction coefficient. They have a definite thermal stability. They have a definite thermal instability. They have a definite thermal resistance. They have a definite thermal conductance. They have a definite thermal capacitance. They have a definite thermal inductance. They have a definite thermal resistance. They have a definite thermal conductance. They have a definite thermal capacitance. They have a definite thermal inductance.

Gas State: Particles are far apart. They have very weak forces of attraction between them. They move very freely. They have no definite volume. They have no definite shape. They have no definite density. They have no definite refractive index. They have no definite thermal conductivity. They have no definite electrical conductivity. They have no definite magnetic permeability. They have no definite optical transparency. They have no definite acoustic transparency. They have no definite thermal expansion coefficient. They have no definite thermal contraction coefficient. They have no definite thermal stability. They have no definite thermal instability. They have no definite thermal resistance. They have no definite thermal conductance. They have no definite thermal capacitance. They have no definite thermal inductance. They have no definite thermal resistance. They have no definite thermal conductance. They have no definite thermal capacitance. They have no definite thermal inductance.



Particle Theory and States of Matter
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Smyth Baden-Powell, Arthur Cowper
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