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 $3\text{Ca(OH)}_2 + 1\text{Al}_2\text{SO}_4 \rightarrow 3\text{CaSO}_4 + 2\text{Al(OH)}_3$ type of reaction double displacement
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 $2\text{Fe} + 3\text{MgO} \rightarrow 2\text{FeO} + 3\text{Mg}$ type of reaction double displacement
 $4\text{C}_2\text{H}_6 + 14\text{O}_2 \rightarrow 8\text{CO}_2 + 12\text{H}_2\text{O}$ type of reaction single displacement

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 $\text{NaOH} + \text{KNO}_3 \rightarrow \text{NaNO}_3 + \text{KOH}$ type of reaction double displacement
 $2\text{Fe} + 6\text{NaBr} \rightarrow 2\text{FeBr}_3 + 6\text{Na}$ type of reaction single displacement
 $\text{MgSO}_4 + 2\text{NH}_4\text{OH} \rightarrow \text{Mg(OH)}_2 + (\text{NH}_4)_2\text{SO}_4$ type of reaction double displacement
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