



CHEMICAL BONDS



IONIC, COVALENT, & METALLIC

FORMS OF CHEMICAL BONDS
Reading comprehension

Students will be able to identify and explain the difference between ionic, covalent, and metallic bonds. They will also be able to identify the types of chemical bonds that form between different elements.

Types of Chemical Bonds

- Ionic:** Formed between a metal and a nonmetal. The metal atom loses one or more electrons, and the nonmetal atom gains one or more electrons. The resulting ions are attracted to each other by electrostatic forces.
- Covalent:** Formed between two nonmetals. The atoms share one or more pairs of electrons, which hold them together.
- Metallic:** Formed between metal atoms. The metal atoms are held together by a "sea" of delocalized electrons.

COVALENT, IONIC, AND METALLIC BONDS

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SCIENCE REAL

A diagram of the periodic table with elements color-coded by groups: metals (shades of blue and green), nonmetals (shades of red and orange), and metalloids (shades of yellow and green).

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A diagram showing the formation of an ionic bond. It starts with a sodium atom (Na) and a chlorine atom (Cl). The sodium atom loses an electron to become a sodium ion (Na+), and the chlorine atom gains that electron to become a chloride ion (Cl-). The two ions are then shown attracted to each other by electrostatic forces.

GUIDED READING ACTIVITY

Ionic Bonds Guided Reading And Study Answers

H Kauffman



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