FORMULAS FOR IONIC COMPOUNDS

1. Write the chemical formula for the ionic compounds made from the following elements.

Elements	metal ion	ion-netal	Chemical Formula	Name	
potassium and iodine	к"	J-1	KI	Portassum lodide	
barium and chlorine	Ba	CI	Ball	Barium Chloride	
sodium and oxygen	NO "+	02-	Na2O	sodium Oxide	
aluminum and nitrogen	AH3"	N,	AIN	Alumina no tride	
lithium and sulfur	L3 '*	St	Ling.	District outide	
magnesium and oxygen	H9*2*	0*	MAD	Magnesium Oxide	
calcium and chlorine	Casa	CITO	Caclz	calcium enloride	
boron and chlorine	8 34	C1'-	Belg	boron enloride	
beryllium and sulfur	Bez+	52.	Bes	beryllium sucker	

2. Complete the chart below and write the ionic formulas to complete it. A few have been done for you.

	F	Cr	02-	S3-	N3-	Бу.
Li*	L: F	E: CI	L;20	Lizs	LizN	Liza
Na"	Nc F	Nacl	Na ₂ 0	Na_s	NasN	Nap
к'	KF	KCI	K ₂ O	K _z S	K, N	K ₂ P
Be ^{2*}	BeFz	BECIZ	BeO	Bes	Be, S.	BezP2
Mg"	MSFz	MgCIz	mgo	MgS	MS3NZ	mg, P,
Ca ²	CaFe	Callz	Cao	Cas	Ca ₂ N ₂	CarPz
A.13*	AIFT	AICI.	41203	Q92 53	AIN	AIP
Ga ³⁴	CIGF3	Gaciz	Gaz01	Gaz Ss	Cran	Gal
B ₂ -	BF,	84,	B203	B 2 53	BN	BP

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