

WORKSHEET – Naming Compounds KEY

Part I – Naming Compounds and Numbering Atoms

FORMULA	NAME	# of DIFFERENT ATOMS	TOTAL # OF ATOMS
1. CaCl_2	Calcium chloride	1 Ca; 2 Cl	3
2. AgCl	<i>Silver chloride</i>	1 Ag; 1 Cl	2
3. MgO	<i>Magnesium oxide</i>	1 Mg; 1 O	2
4. NaBr	<i>Sodium bromide</i>	1 Na; 1 Br	2
5. Al_2O_3	<i>Aluminum oxide</i>	2 Al; 3 O	5
6. KI	<i>Potassium iodide</i>	1 K; 1 I	2
7. ZnCl_2	<i>Zinc chloride</i>	1 Zn; 2 Cl	3
8. Li_2O	<i>Lithium oxide</i>	2 Li; 1 O	3
9. BaO	<i>Barium oxide</i>	1 Ba; 1 O	2
10. CaBr_2	<i>Calcium bromide</i>	1 Ca; 2 Br	3
11. MgCl_2	<i>Magnesium chloride</i>	1 Mg; 2 Cl	3
12. AgI	<i>Silver iodide</i>	1 Ag; 1 I	2
13. ZnS	<i>Zinc sulphide</i>	1 Zn; 1 S	2
14. BaF_2	<i>Barium fluoride</i>	1 Ba; 2 F	3
15. Ca_3P_2	<i>Calcium phosphide</i>	3 Ca; 2 P	5
16. Na_2O	<i>Sodium oxide</i>	2 Na; 1 O	3
17. AlN	<i>Aluminum nitride</i>	1 Al; 1 N	2
18. K_3N	<i>Potassium nitride</i>	3 K; 1 N	4
19. Ca_2C	<i>Calcium carbide</i>	2 Ca; 1 C	3
20. K_2O	<i>Potassium oxide</i>	2 K; 1 O	3

Ionic Compounds Homework Review Sheet Answer Key

G Thomas



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Foundations of College Chemistry Morris Hein, Susan Arena, Cary Willard, 2016-08-02 This text is an unbound three hole punched version Used by over 750 000 students Foundations of College Chemistry Binder Ready Version 15th Edition is praised for its accuracy clear no nonsense approach and direct writing style Foundations direct and straightforward explanations focus on problem solving making it the most dependable text on the market Its comprehensive scope proven track record outstanding in text examples and problem sets were all designed to provide instructors with a solid text while not overwhelming students in a difficult course Foundations fits into the prep intro chemistry courses which often include a wide mix of students from science majors not yet ready for general chemistry allied health students in their 1st semester of a GOB sequence science education students for elementary school teachers to the occasional liberal arts student fulfilling a science requirement Foundations was specifically designed to meet this wide array of needs **Chemistry, Student Study**

Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement **Investigating Chemistry** Matthew Johll, 2008-12-22 In its new second edition Investigating Chemistry A

Forensic Science Perspective remains the only book that uses the inherently fascinating topics of crime and criminal investigations as a context for teaching the fundamental chemical concepts most often covered in an introductory nonmajors course Covering all the standard topics Matthew Johll capitalizes on the surge of interest in the scientific investigation of crime as sparked by CSI and other television shows bringing together the theme of forensic science and the fundamentals of chemistry in ways that are effective and accessible for students This edition features refined explanations of the chemical concepts which are the core of the book as well as a more thoroughly integrated forensic theme updated features and an expanded media supplements package **Holt Chemistry** Ralph Thomas Myers, 2004 **Science Software Quarterly**

, 1985 **Books in Print Supplement**, 2002 **Science Tests and Reviews** Buros Center, 1975 Science Tests and Reviews consisting of science sections of the first seven MMYs and Tests in Print II includes 217 original test reviews written by 81 specialists 18 excerpted test reviews 270 references on the construction use and validity of specific tests a bibliography on in print science tests references for specific tests cumulative name indexes for specific tests with references a publishers directory title index name index and a scanning index The 97 tests covered fall into the following categories 23 general 14

biology 35 chemistry 3 geology 6 miscellaneous and 16 physics *Foundations of Physical Science* Tom Hsu PH.D.,2002

Chapter Resource 2 Chemistry of Life Biology Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2004

Merrill Chemistry Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998 **The ... Mental Measurements**

Yearbook Oscar Krisen Buros,1941 *Ceramic Abstracts* American Ceramic Society,1981 *A Level Chemistry Multiple Choice Questions and Answers (MCQs)* Arshad Iqbal,2020-04-10 Previously published as A Level Chemistry MCQs Multiple Choice Questions and Answers Quiz Tests with Answer Keys by Arshad Iqbal A Level Chemistry Multiple Choice Questions and Answers MCQs A Level Chemistry quizzes practice tests with answer key provides mock tests for competitive exams to solve 1745 MCQs A Level Chemistry MCQs helps with theoretical conceptual and analytical study for self assessment career tests This book can help to learn and practice A Level Chemistry quizzes as a quick study guide for placement test preparation A level Chemistry Multiple Choice Questions and Answers MCQs is a revision guide with a collection of trivia quiz questions and answers on topics Alcohols and esters atomic structure and theory benzene chemical compound carbonyl compounds carboxylic acids acyl compounds chemical bonding chemistry of life electrode potential electrons in atoms enthalpy change equilibrium group IV groups II and VII halogenoalkanes hydrocarbons introduction to organic chemistry ionic equilibria lattice energy moles and equations nitrogen and sulfur organic and nitrogen compounds periodicity polymerization rates of reaction reaction kinetics redox reactions and electrolysis states of matter transition elements to enhance teaching and learning A level Chemistry Quiz Questions and Answers also covers the syllabus of many competitive papers for admission exams of different universities from chemistry textbooks on chapters Alcohols and Esters Multiple Choice Questions 27 MCQs Atomic Structure and Theory Multiple Choice Questions 37 MCQs Benzene Chemical Compound Multiple Choice Questions 41 MCQs Carbonyl Compounds Multiple Choice Questions 29 MCQs Carboxylic Acids and Acyl Compounds Multiple Choice Questions 27 MCQs Chemical Bonding Multiple Choice Questions 213 MCQs Chemistry of Life Multiple Choice Questions 29 MCQs Electrode Potential Multiple Choice Questions 62 MCQs Electrons in Atoms Multiple Choice Questions 53 MCQs Enthalpy Change Multiple Choice Questions 45 MCQs Equilibrium Multiple Choice Questions 50 MCQs Group IV Multiple Choice Questions 53 MCQs Groups II and VII Multiple Choice Questions 180 MCQs Halogenoalkanes Multiple Choice Questions 33 MCQs Hydrocarbons Multiple Choice Questions 53 MCQs Introduction to Organic Chemistry Multiple Choice Questions 52 MCQs Ionic Equilibria Multiple Choice Questions 56 MCQs Lattice Energy Multiple Choice Questions 33 MCQs Moles and Equations Multiple Choice Questions 50 MCQs Nitrogen and Sulfur Multiple Choice Questions 89 MCQs Organic and Nitrogen Compounds Multiple Choice Questions 54 MCQs Periodicity Multiple Choice Questions 202 MCQs Polymerization Multiple Choice Questions 36 MCQs Rates of Reaction Multiple Choice Questions 39 MCQs Reaction Kinetics Multiple Choice Questions 52 MCQs Redox Reactions and Electrolysis Multiple Choice Questions 55 MCQs States of Matter Multiple Choice Questions 66 MCQs Transition Elements Multiple Choice Questions 29

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