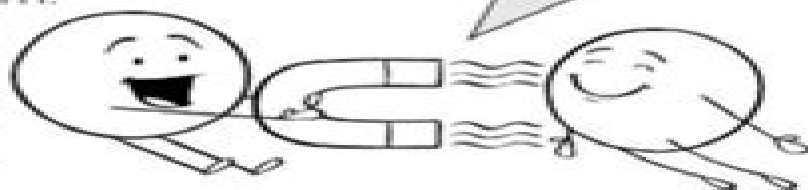


CHEMICAL BONDS

Ionic Bonding

Metals and non-metals react together to form _____ compounds. They are held together by ionic bonds. Ionic bonding is a force of _____ attraction between two _____ atoms (called _____). An atom becomes charged by either donating or accepting _____ electrons from another atom. A valence electron is one from the _____ shell of the atom.

Since opposites attract, there is an attraction between the positively charged and negatively charged ions that holds them together.

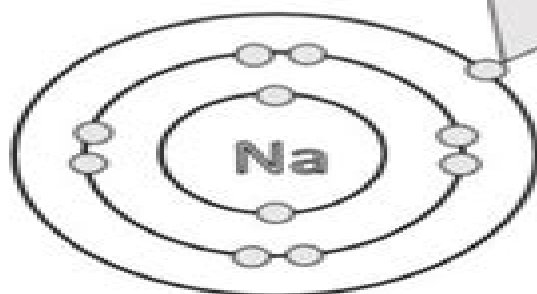


If an atom _____ an electron, it becomes positively charged. This is then called a _____. _____ always donate electrons. If an atom accepts an electron, it becomes _____. This is called an _____. Non-metals always accept electrons.

Atoms do this in order to become '_____'. A stable atom is one which has obtained a _____ of electrons.

How do we know which atoms donate and which accept electrons?

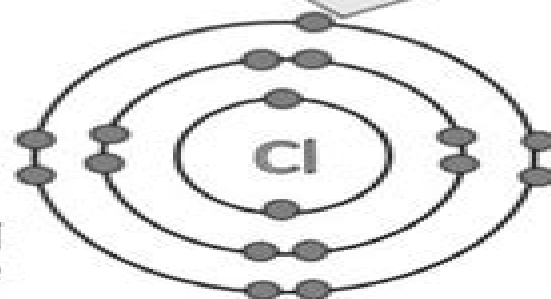
Sodium has _____ lone electron in the outer shell. It is easiest for this atom to _____ one electron than to try and find seven more to fill the outer shell.



All _____ donate electrons and therefore become cations.

All _____ accept electrons and therefore become anions.

Chlorine has _____ electrons in its outer shell. Therefore, it is easiest for Chlorine to _____ one electron than to _____ to try and donate seven.



Ionic Bond Middle School Worksheet

Margaret Irene Lindsay

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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Atoms Electrons Structure and Bonding Workbook

Topics Covered are

- Some of this content has also been published in the Summer Start to A Level Chemistry and a practice exam paper
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- Reference table of common ions formulae
- Formula of Ionic Compounds 65 formulas to work out
- Drawing Ionic Bonding 10 Compounds
- Simple Covalent Bonding 20 multiple choice questions
- Drawing Covalent Bonding 10 Compounds
- Summary Table for the 4 Different Types of Bonding
- Electron configurations
- Drawing electron configurations
- Drawing Electron Configurations Spot the mistake
- Electronic Configuration 20 multiple choice questions
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- Balancing Equations using the oxidation numbers method 20 to practice
- Salt Equations 20 equations to complete and balance
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Table of Contents Ionic Bond Middle School Worksheet

1. Understanding the eBook Ionic Bond Middle School Worksheet
 - The Rise of Digital Reading Ionic Bond Middle School Worksheet
 - Advantages of eBooks Over Traditional Books
2. Identifying Ionic Bond Middle School Worksheet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ionic Bond Middle School Worksheet
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ionic Bond Middle School Worksheet
 - Personalized Recommendations
 - Ionic Bond Middle School Worksheet User Reviews and Ratings
 - Ionic Bond Middle School Worksheet and Bestseller Lists
5. Accessing Ionic Bond Middle School Worksheet Free and Paid eBooks
 - Ionic Bond Middle School Worksheet Public Domain eBooks
 - Ionic Bond Middle School Worksheet eBook Subscription Services
 - Ionic Bond Middle School Worksheet Budget-Friendly Options

6. Navigating Ionic Bond Middle School Worksheet eBook Formats
 - ePub, PDF, MOBI, and More
 - Ionic Bond Middle School Worksheet Compatibility with Devices
 - Ionic Bond Middle School Worksheet Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ionic Bond Middle School Worksheet
 - Highlighting and Note-Taking Ionic Bond Middle School Worksheet
 - Interactive Elements Ionic Bond Middle School Worksheet
8. Staying Engaged with Ionic Bond Middle School Worksheet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ionic Bond Middle School Worksheet
9. Balancing eBooks and Physical Books Ionic Bond Middle School Worksheet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ionic Bond Middle School Worksheet
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ionic Bond Middle School Worksheet
 - Setting Reading Goals Ionic Bond Middle School Worksheet
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ionic Bond Middle School Worksheet
 - Fact-Checking eBook Content of Ionic Bond Middle School Worksheet
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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