

ATOMIC STRUCTURE

RUTHERFORD'S MODEL

- Discovered that an atom is mostly made up of empty space
- Center of the atom is a small dense positively charged nucleus
- This nucleus holds the negatively charged particles in the atom

BOHR'S MODEL

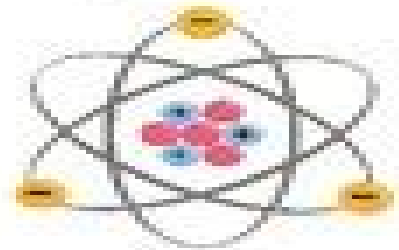
- There are 2 main pieces of an atom:
- Nucleus
- Electrons

THE OLD MODEL OF THE ATOM



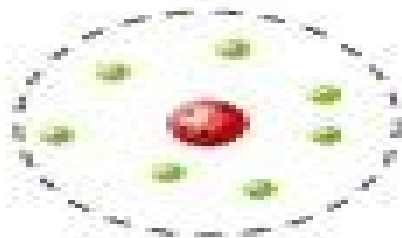
WAVE-MECHANICAL MODEL

- Electrons are moving so fast that they look like a cloud. Orbitals are regions of most likely electron location



WHAT IS TRUE TODAY...

- The nucleus is composed of 2 things:
 - PROTONS - positively charged particles
 - NEUTRONS - neutral charged particles
- Electrons are in a cloud - they are in orbitals
 - Electrons DO move around the nucleus in different patterns



Subatomic Particle	Location in the Atom	Charge	Weight
proton	center (in the nucleus)	positive	1 amu
neutron	center (in the nucleus)	neutral (no charge)	1 amu
electron	around the nucleus in clouds or orbitals	negative	Almost equal to neutrons (1/2)

IONS

- If an atom gains or loses an electron, it is called an ion
- This gives the atom a charge:
 - Lose an electron, the atom becomes positive
 - Gain an electron, the atom becomes negative

ISOTOPES

- An isotope is when an atom is physically different
- An isotope is still the same element, it is just a little different from every other atom of the same element
 - slightly different mass

Lifepac Chemistry Atomic Structure Teacher Guide

Michael P. Olmsted



Lifepac Chemistry Atomic Structure Teacher Guide:

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Congress. House. Committee on Small Business. Subcommittee on Regulation, Business Opportunities, and Energy,1989

Explaining Chemical Change : Student Exercises and Teacher's Guide for Grade Nine Academic Science Jim (James William) Ross,Jim Ross,Mike Lattner,2004 Take your beginning chemistry students farther than you ever imagined The Ross Lattner Intuitive Chemistry books start with the traditional chemistry curriculum and modify it in three unique ways 1 Uses the Ross model of the atom a simplified version of the atom that works the way that young people think Students can use the Ross model to make insightful explanations and accurate predictions of chemical behaviour 2 Rearranges the traditional curriculum to avoid confusion and build student competency Competent students are confident students 3 Provides a model of science that directs the student s attention to the student s own understanding Each experiment tests the validity of the student s own thinking As in all of the Ross Lattner IntuitivScience series diagrams are an important mode of expression Parents and teachers you get one half of the book We provide solid pedagogical supports recipes and methods of presentation Informed by over thirty years of international research into how students learn science the IntuitivChemistry series directly challenges the everyday misconceptions that students bring to your classroom **Big Book of Home Learning** Mary Pride,1991-07 Learn at home with exciting products for all school subjects New **Atomic Structure** Sammie Jacobs,2020-06-24 Teachers parents and educators of all kinds welcome to my classroom Here within these pages and in fact this entire series are lessons and activities that I have created and used with my own students over the years presented in an easy to use format Through trial and error hardship and success I have learned how to present the difficult world of introductory chemistry to students in scalable terms ways that lend themselves to a wide variety of learners So if you teach in a small class large class average class have ESL or SPED or even if you are homeschooling you will find the flexibility of my lesson plans tailor made for you Each unit includes a complete calendar plan lessons for each classroom day

for forty five minutes of instruction each and even formative and summative assessments to check for learning Simply use the lessons scale them as need be and augment whenever you like Be sure to join my Teaching Chemistry PLC on Facebook for even more ideas and sharing and of course I hope for the best with you and your students Sammie Jacobs

Chemistry (Teacher Guide) Dr. Dennis Englin, 2018-02-26 This book was created to help teachers as they instruct students through the Master's Class Chemistry course by Master Books The teacher is one who guides students through the subject matter helps each student stay on schedule and be organized and is their source of accountability along the way With that in mind this guide provides additional help through the laboratory exercises as well as lessons quizzes and examinations that are provided along with the answers The lessons in this study emphasize working through procedures and problem solving by learning patterns The vocabulary is kept at the essential level Practice exercises are given with their answers so that the patterns can be used in problem solving These lessons and laboratory exercises are the result of over 30 years of teaching home school high school students and then working with them as they proceed through college Guided labs are provided to enhance instruction of weekly lessons There are many principles and truths given to us in Scripture by the God that created the universe and all of the laws by which it functions It is important to see the hand of God and His principles and wisdom as it plays out in chemistry This course integrates what God has told us in the context of this study Features Each suggested weekly schedule has five easy to manage lessons that combine reading and worksheets Worksheets quizzes and tests are perforated and three hole punched materials are easy to tear out hand out grade and store Adjust the schedule and materials needed to best work within your educational program Space is given for assignments dates There is flexibility in scheduling Adapt the days to your school schedule Workflow Students will read the pages in their book and then complete each section of the teacher guide They should be encouraged to complete as many of the activities and projects as possible as well Tests are given at regular intervals with space to record each grade About the Author DR DENNIS ENGLIN earned his bachelor's from Westmont College his master of science from California State University and his EdD from the University of Southern California He enjoys teaching animal biology vertebrate biology wildlife biology organismic biology and astronomy at The Master's University His professional memberships include the Creation Research Society the American Fisheries Association Southern California Academy of Sciences Yellowstone Association and Au Sable Institute of Environmental Studies

Chemistry Joseph Castka, 1962 **Reconstructing Atomic Theory**, 2009-01-01 Inquiries in Science Chemistry Series Reconstructing Atomic Theory Teacher's Guide ATOMIC STRUCTURE High School Chemistry Curriculum Bud Morgan, 2025-07-09 Welcome to our comprehensive guide to Atomic Structure designed specifically for high school students studying chemistry In this book we will explore the fascinating world of atoms the basic building blocks of matter and delve into the complexities of their structure Chapter 1 provides a solid foundation by introducing the concept of atoms and their historical significance in the development of modern chemistry We will examine the key scientists who contributed to our

understanding of atomic structure from Democritus to Thomson Rutherford and Bohr Moving on to Chapter 2 we will explore the structure of the atom in more detail including the subatomic particles that make up an atom protons neutrons and electrons We will also discuss the concept of atomic number atomic mass and isotopes In Chapter 3 we will delve into the arrangement of electrons within an atom s electron cloud including energy levels sublevels and orbital shapes We will also discuss the principles of electron configuration and how it impacts an atom s chemical properties Chapter 4 will focus on the periodic table and how it organizes elements based on their atomic structure We will explore periodic trends such as atomic radius ionization energy and electron affinity and discuss how these trends are related to an element s position on the table In our final chapter we will discuss the significance of atomic structure in various chemical reactions including bonding stoichiometry and the behavior of gases We will also explore the importance of understanding atomic structure in fields such as biochemistry environmental science and materials science Throughout this book we will provide clear explanations helpful illustrations and interactive activities to reinforce key concepts Whether you are a visual learner a hands on learner or a theoretical learner this book is designed to meet your needs We hope that this book will not only enhance your understanding of atomic structure but also inspire you to further explore the wonders of chemistry So let s dive in and unlock the mysteries of the atom together

Nuffield Physics ,1971 *Chemistry Models - Atoms and the Periodic Table: Teacher's Guide* ,2013 *Chemistry 1* Donald A. Cook,Joseph F. Castka,1962 **Chemistry Teacher's Guide** Michael P. Olmsted,1972

Chemistry I Basic Systems, Inc,1962 *Chemistry Lesson Plans, Study Guides, and Lecture Notes* Doctor Chemistry,2017-08-13 The Ultimate Guide to Learning or Teaching Chemistry This book contains the real lecture notes and slide of a highly effective high school and college Chemistry teacher Teachers Never plan another lesson again Students Ace your upcoming exam This series covers all of the topics of High School Chemistry and General Chemistry including Accuracy and Significant FiguresMixtures Metric SystemBondingAtomic TheoryPeriodic TableVSEPRIonic and Covalent BondingGeometric BondingThe Mole and Molar MassEquation BalancingThermodynamicsStoichiometryStates of MatterGas Laws and CalculationsReaction CalculationsAcids and BasesLimiting ReagentsRedox and Electro ChemistryOrganic Chemistry Basics *Properties of Atoms & Molecules Teacher Supplement* Debbie &. Richard Lawrence,2016-02-01 This teacher supplement book provides an introduction on how to teach the curriculum a supply list and answer key for each lesson a resource guide containing suggested books videos and field trips and a master supply list for God s Design for Chemistry and Ecology Properties of Atoms and Molecules Also includes student supplement worksheets and tests in an electronic form *Atomic Structure* Nuffield Foundation,1979

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