

- (vi) Some candidates incorrectly answered as efflorescent or hygroscopic.
- (g) (i) Common error that was repeated among all the sub parts was that even if the reagent was correctly chosen the result with both the substances under consideration was not specified.  
The reagent, conc.  $\text{H}_2\text{SO}_4$ , was incorrectly chosen by many candidates.  
The solution state of  $\text{NaCl}$  and  $\text{NaNO}_3$ .
- (ii) Many candidates differentiated on the basis of smell and did not fulfil the requirement of chemical test. Some suggested blue litmus turning red for  $\text{HCl}$  and the other way round for  $\text{H}_2\text{S}$ . Few of the candidates did not use the same reagent for both gases.
- (iii) Many candidates brought out the difference on the basis of bonds present i.e. double bond and single bonds and did not give a chemical test to distinguish. Some other candidates specified the kind of reactions shown by both i.e. addition and substitution reactions respectively or the type of compounds as being unsaturated and saturated.
- (iv) Candidates used  $\text{NaOH}$  as the distinguishing reagent and stated the colour of the ppt with each solution but failed to specify the solubilities of the ppt in the reagent.  
Many candidates were not aware that  $\text{Ca}(\text{NO}_3)_2$  solution does not form ppt with  $\text{NH}_4\text{OH}$ .
- (v) Many candidates did not associate the word solution / paper / acidified with  $\text{K}_2\text{Cr}_2\text{O}_7$  or  $\text{KMnO}_4$ .  
Some others used lime water test which was answered by both gases.
- (f) (i) Chlorine and lithium were the occasional wrong answers.
- (ii) Most attempted correctly. Few candidates chose Ammonia as the correct options.
- (iii) Most candidates answered correctly.
- (iv) Some of the candidates wrote acetic acid.
- (v) Most of the candidates answered correctly.
- (vi) Oxide chosen at random.
- (vii) Some incorrectly answered this part as dehydrating agent.
- (viii) Generally correctly answered. Few candidates incorrectly answered as Ammonium nitrate.
- (ix) Most candidates answered this part correctly. Occasional errors showed a lack of understanding of the properties of ionic compounds.
- (x) There was confusion between the components of brass and bronze.
- (g) (i) Candidates failed to calculate the gram molecular weight correctly.
- (ii) Knowledge of gram molecular volume was missing.
- (iii) Most candidates answered this part correctly. Some suggested  $6.023 \times 10^{23}$  molecules of  $\text{N}_2$  or multiplied  $\text{N}$  by  $6.023 \times 10^{23}$  molecules.

#### Suggestions for teachers

- Students must be clear about the basic concepts so that understanding relation between mole, molar mass and molar volume is possible.
- Students must be given adequate practice in application of Avogadro's law after explaining it and pointing out the variations from Gay Lussac's law

# Leaked 2014 Icse Paper 1accounting

**Roman Wölfel**



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## **Table of Contents Leaked 2014 Icse Paper 1accounting**

1. Understanding the eBook Leaked 2014 Icse Paper 1accounting
  - The Rise of Digital Reading Leaked 2014 Icse Paper 1accounting
  - Advantages of eBooks Over Traditional Books
2. Identifying Leaked 2014 Icse Paper 1accounting
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Leaked 2014 Icse Paper 1accounting
  - User-Friendly Interface
4. Exploring eBook Recommendations from Leaked 2014 Icse Paper 1accounting
  - Personalized Recommendations
  - Leaked 2014 Icse Paper 1accounting User Reviews and Ratings
  - Leaked 2014 Icse Paper 1accounting and Bestseller Lists
5. Accessing Leaked 2014 Icse Paper 1accounting Free and Paid eBooks
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  - Leaked 2014 Icse Paper 1accounting eBook Subscription Services
  - Leaked 2014 Icse Paper 1accounting Budget-Friendly Options
6. Navigating Leaked 2014 Icse Paper 1accounting eBook Formats
  - ePub, PDF, MOBI, and More
  - Leaked 2014 Icse Paper 1accounting Compatibility with Devices
  - Leaked 2014 Icse Paper 1accounting Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Leaked 2014 Icse Paper 1accounting
  - Highlighting and Note-Taking Leaked 2014 Icse Paper 1accounting
  - Interactive Elements Leaked 2014 Icse Paper 1accounting

8. Staying Engaged with Leaked 2014 Icse Paper 1accounting
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Leaked 2014 Icse Paper 1accounting
9. Balancing eBooks and Physical Books Leaked 2014 Icse Paper 1accounting
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Leaked 2014 Icse Paper 1accounting
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Leaked 2014 Icse Paper 1accounting
  - Setting Reading Goals Leaked 2014 Icse Paper 1accounting
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Leaked 2014 Icse Paper 1accounting
  - Fact-Checking eBook Content of Leaked 2014 Icse Paper 1accounting
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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