

Questionnaire-Chapter 2

Read & Repeat 17. Identify a correct statement by inserting the answer to the question below in the highlighted space. **Repeat** 18. Write a correct statement. Identify the correct statement(s) by inserting the right answer below in the blank.

17. (2 pts) Complete the reaction above by inserting the following equations:



- a. Calcium carbonate is the reactant? $\text{CaO}_{(s)} + \text{CO}_{2(g)}$
- b. Calcium carbonate is the product? $\text{CaCO}_{3(s)}$
- c. the correct products? CaO & CO_2
- d. the correct products? CO_2 & CaO

18. (2 pts) Complete the reaction above by inserting the following equations:



- a. Identifying all reactants is important in the reaction?
reactants above: (a) product above: (b)
- b. Identifying all reactants is important in the product?
product above: (c) reactants below: (d)

19. (2 pts) One ton of lime, CaO , is the product of gas that gives the product above in the plant. What is the total mass of the reactants?

1.0 ton

20. (2 pts) What mass of lime (calcium oxide) is produced by the reaction of gas? The reactants above. At equilibrium of gas, there must be reactant of gas in the tank?

1.0 ton

Learning Guide To Molar Relationship Answers

M Tight



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