

Name: \_\_\_\_\_ Date: \_\_\_\_\_ Class: \_\_\_\_\_

## What do You Know About Acids and Bases?

Letter	Matching	
	1. Acid	A. To mix acids and bases to cancel each other out and make water and salt
	2. Base	B. A compound that adds $H^+$ ions to water
	3. Neutral	C. Equal number of $H^+$ and $OH^-$ ions; water is an example
	4. Neutralize	D. A compound that adds $OH^-$ ions to water.
	5. Acid Rain	E. When pollution causes rain to have a pH less than 5.6.
	6. pH	F. The measure of how acidic or basic a solution is
	7. Salt Water	G. A compound that adds a few $OH^-$ ions to water
	8. Strong Acid	H. The product of a neutralization reaction between an acid and a base
	9. Weak Base	I. A compound that adds a few $H^+$ ions to water
	10. Weak Acid	J. A compound that adds a lot of $H^+$ ions to water.

Is it an Acid or a Base?		
Circle the acids and underline the bases		
11. HCl	17. $H_2CO_3$	23. apple juice
12. $Mg(OH)_2$	18. NaOH	24. lemonade
13. $H_3PO_4$	19. $Al(OH)_3$	25. soap
14. KOH	20. HBr	26. laundry detergent
15. $Ca(OH)_2$	21. $H_2SO_4$	27. soft drinks
16. LiOH	22. $H_2O$	28. bathroom cleaner

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**Robert A. Richardson**



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