

Acids and Bases

How do acids and bases behave in water?

Why?

Acids and bases play an important role in our lives. Numerous biological processes, industrial applications, and even environmental problems are a function of the acidity or basicity (alkalinity) of aqueous solutions. It is therefore important to understand what makes a substance behave as an acid or a base when dissolved in water. In this activity, we will explore the physical and chemical properties of acids and bases.

Model 1 – Arrhenius Acids and Bases

Common Name for Aqueous Solution	Chemical Formula	Found in...	Tastes...	Turns Litmus Paper...	Conducts Electricity?	Acid or Base
Acetic acid	$\text{HC}_2\text{H}_3\text{O}_2(\text{aq})$	Vinegar	Sour	Red	Yes	Acid
Benzoic acid	$\text{HC}_7\text{H}_5\text{O}_2(\text{aq})$	Food preservative	Sour	Red	Yes	Acid
Phosphoric acid	$\text{H}_3\text{PO}_4(\text{aq})$	Soda pop	Sour	Red	Yes	Acid
Hydrochloric acid	$\text{HCl}(\text{aq})$	Stomach acid	Sour	Red	Yes	Acid
Citric acid	$\text{H}_2\text{C}_6\text{H}_7\text{O}_6(\text{aq})$	Citrus fruits	Sour	Red	Yes	Acid
Ascorbic acid	$\text{H}_2\text{C}_6\text{H}_7\text{O}_6(\text{aq})$	Vitamin C	Sour	Red	Yes	Acid
Magnesium hydroxide	$\text{Mg}(\text{OH})_2(\text{aq})$	Milk of magnesia	Bitter	Blue	Yes	Base
Aluminum hydroxide	$\text{Al}(\text{OH})_3(\text{aq})$	Antacids	Bitter	Blue	Yes	Base
Barium hydroxide	$\text{Ba}(\text{OH})_2(\text{aq})$	Lubricants	POISON	Blue	Yes	Base
Sodium hydroxide	$\text{NaOH}(\text{aq})$	Drain cleaner	POISON	Blue	Yes	Base

- Refer to Model 1.
 - What is the common chemical name for Vitamin C?
 - Is Vitamin C classified as an acid or a base?
- Examine the properties of the Arrhenius acids in Model 1. List three properties that all Arrhenius acids have in common.
- Examine the chemical formulas for the Arrhenius acids in Model 1. What feature do all the Arrhenius acid chemical formulas have in common?
- Examine the properties of the Arrhenius bases in Model 1. List two properties that all Arrhenius bases have in common.

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Emil Abderhalden



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Table of Contents Lecture 32 Acids And Bases Iv Answers

1. Understanding the eBook Lecture 32 Acids And Bases Iv Answers
 - The Rise of Digital Reading Lecture 32 Acids And Bases Iv Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Lecture 32 Acids And Bases Iv Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Lecture 32 Acids And Bases Iv Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Lecture 32 Acids And Bases Iv Answers
 - Personalized Recommendations
 - Lecture 32 Acids And Bases Iv Answers User Reviews and Ratings
 - Lecture 32 Acids And Bases Iv Answers and Bestseller Lists
5. Accessing Lecture 32 Acids And Bases Iv Answers Free and Paid eBooks
 - Lecture 32 Acids And Bases Iv Answers Public Domain eBooks
 - Lecture 32 Acids And Bases Iv Answers eBook Subscription Services

- Lecture 32 Acids And Bases Iv Answers Budget-Friendly Options
- 6. Navigating Lecture 32 Acids And Bases Iv Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Lecture 32 Acids And Bases Iv Answers Compatibility with Devices
 - Lecture 32 Acids And Bases Iv Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Lecture 32 Acids And Bases Iv Answers
 - Highlighting and Note-Taking Lecture 32 Acids And Bases Iv Answers
 - Interactive Elements Lecture 32 Acids And Bases Iv Answers
- 8. Staying Engaged with Lecture 32 Acids And Bases Iv Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Lecture 32 Acids And Bases Iv Answers
- 9. Balancing eBooks and Physical Books Lecture 32 Acids And Bases Iv Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Lecture 32 Acids And Bases Iv Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Lecture 32 Acids And Bases Iv Answers
 - Setting Reading Goals Lecture 32 Acids And Bases Iv Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Lecture 32 Acids And Bases Iv Answers
 - Fact-Checking eBook Content of Lecture 32 Acids And Bases Iv Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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