



# Leica M Manual

**SJ Ball**



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web in this article we will discuss polyatomic ions the prefix poly means many so a polyatomic ion is an ion that contains more than one atom this differentiates polyatomic ions from monatomic ions which contain only one atom examples of monatomic ions include na fe 3 cl and many many others

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web because these ions contain more than one atom they are called polyatomic ions polyatomic ions have characteristic formulas names and charges that should be memorized for example the nitrate ion has one nitrogen atom and three oxygen atoms with an overall charge of 1

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web aug 23 2019 these are called polyatomic ions table 2 lists the formulas charges and names of some common polyatomic ions only one of them the ammonium ion is a cation the rest are anions most of them also contain oxygen atoms so sometimes they are referred to as oxyanions

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nitrite perphosphate hypoiodite chlorite phosphite percarbonate bromate hyposulfite permanganate carbonite  $\text{co}_3^{2-}$   $\text{mno}_3^-$   $\text{so}_5^{2-}$   $\text{bro}_2^-$   $\text{co}_4^{2-}$   $\text{po}_5^{3-}$   $\text{clo}_1^-$   $\text{io}_2^-$   $\text{po}_4^{3-}$   $\text{no}_2^-$   $\text{so}_4^{2-}$

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web polyatomic ions are charged groups of atoms an example is ammonium ion  $\text{nh}_4^+$  it has five atoms one nitrogen and four hydrogens that share a charge of 1 the polyatomic ions remain intact and parentheses may be required when using subscripts for example ammonium chloride is  $\text{nh}_4\text{cl}$  and ammonium sulfide is  $\text{nh}_4_2\text{s}$

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web polyatomic ions have defined formulas names and charges that cannot be modified in any way table 3 4 1 3 4 1 lists the ion names and ion formulas of the most common polyatomic ions for example  $\text{no}_3^-$  is the nitrate ion it has one nitrogen atom and three oxygen atoms and an overall 1 charge

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web rules for naming ionic compounds containing polyatomic ions polyatomic ions are ions which consist of more than one atom for example nitrate ion  $\text{no}_3^-$  contains one nitrogen atom and three oxygen atoms the atoms in a polyatomic ion are usually covalently bonded to one another and therefore stay together as a single charged unit

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web ions made up of more than one atom are known as polyatomic ions ions with positive charge are called cations ions with negative charge are called anions list of monatomic ions the ions made of a single atom are called simple ions or monatomic ions frequently asked questions on polyatomic ions list q1 what are 3 examples of polyatomic ions

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polyatomic ion don't add a number to the name

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web nov 27 2018 name of ion charge on ion type and number of atoms chemical formula polyatomic ions nitride i nitrogen nitrate i nitrogen 3 oxygen 2 consider the terms monatomic and polyatomic as they are used in model 1 write a definition for each of these terms it may be helpful to break the words apart i e poly atomic

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web aug 11 2022 to identify and name polyatomic ions some ions consist of groups of atoms covalently bonded together and have an overall electric charge because these ions contain more than one atom they are called polyatomic ions the structures

names and formulas of some polyatomic ions are found in the below figure and table

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web the formulae of compounds containing polyatomic ions are worked out in a similar way to single atom ions except when there is more than one polyatomic ion then its formula is written

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of you

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