Acids and bases

There are three classes of strong electrolytes.

1 Strong Water Soluble Acids

$$HNO_{3(\ell)} + H_2O_{(\ell)} \xrightarrow{\approx 100\%} H_3O_{(aq)}^+ + NO_{3(aq)}^-$$

2 Strong Water Soluble Bases

$$KOH_{(s)} \xrightarrow{H_2O \approx 100\%} K_{(aq)}^+ + OH_{(aq)}^-$$

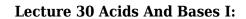
3 Most Water Soluble Salts

$$NaCl_{(s)} \xrightarrow{H_2O \approx 100\%} Na_{(aq)}^+ + Cl_{(aq)}^-$$

 Weak acids and bases ionize or dissociate partially, much less than 100%, and is often less than 10%.

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