

**Potentiometry**

Choose the correct answer:

No	Sentence	answer
1	All of the following are related to reference electrode except: a) of known and constant potential b) obeys Nernst equation. c) of high resistance to permit current of electric current. d) insensitive to activity of analyte solution.	C
2	Potential of standard hydrogen electrode is: a) Zero b) related to $H^+$ concentration c) 0.059 PH d) A and B e) All of the above	d
3	The increase in chloride ion concentration in Ag/AgCl electrode leads to: a) increase the potential b) decrease the potential c) no effect d) none of them	b
4	Potential of calomel electrode is ..... proportional to chloride ion concentration. a) directly b) inversely c) equal d) no relation	b
5	In silver/silver chloride electrode, as KCl concentration increases: a) silver ions increases, potential increases. b) silver ions decreases, potential increases. c) silver ions decreases, potential remains the same. d) silver ions decreases, potential decreases.	d
6	In calomel electrode, as KCl concentration increases: a) mercury(I) ions increases, potential increases. b) mercury(I) ions decreases, potential increases. c) mercury(I) ions decreases, potential remains the same. d) mercury(I) ions decreases, potential decreases.	d
7	What is the electrode potential for a half-cell consisting of a cadmium electrode immersed in a solution that is 0.0150 M in $Cd^{2+}$ ( $E^0 = -0.403$ V) a) 0.457 b) 0.403 c) -0.457 d) -0.201	C

# Mcq On Potentiometry

**Shikha Agarwal**



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Fouletier,Pierre Fabry,2010-01-26 This book reviews the state of art in the field of chemical sensors for analyses of ionic or molecular species dissolved in liquid media mainly in aqueous solutions The transduction of such devices is based on chemical biological and physical phenomena The fundamental phenomena involved in these sensors are described in the different chapters by specialists having a good expertise in the field Numerous recent bibliographic references are given Most of the devices could be miniaturised using modern technologies allowing a fabrication on a large scale for a mass

production at low cost Moreover such devices could open the field of applications in a near future environmental biomedical food industries domestic and automotive applications etc

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**Medical Laboratory Technology Examination Review** Zandra E. Strother, 1985

**Subject Guide to Books in Print**, 1984

Books in Print Supplement, 1988

*Potentiometry and Potentiometric Titrations* E. P. Serjeant, 1984-06-29 Cells electrodes and basic procedures rudiments of potentiometry classifications of potentiometry and titrimetry classifications and descriptions of electrodes procedures of analytical potentiometry electrodes for potentiometric titrimetry Applications to the determination of solution equilibrium data the determination of ionization constants in aqueous solution acid base equilibria in nonaqueous solvents the determination of stability constants by potentiometric titration Applications to titrimetric analysis the basics of titrimetry acid base titrimetry complexometric titrations oxidation reduction titrations precipitation titrations

**Advanced Potentiometry** Erzsébet Néher-Neumann, 2010-03-14 Advanced Potentiometry details the accurate calculation of potentiometric titrations Additional terms such as the ideal diffusion potential together with the contribution of activity factors furnish calculated stability constants which do not include systematic errors and thus present a complete and correct description of equilibrium processes Moreover serious systematic errors resulting from the recent use of curve fitting computer programs are highlighted later in the text Advanced Potentiometry can advantageously be used by lecturers researchers PhD graduates and undergraduate students working in the fields of Solution Chemistry Physical Chemistry Inorganic Chemistry and Analytical Chemistry and Environmental Research

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**Ionizable Compounds in Two-phase Systems** Per-Arne Johansson,1977

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In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Mcq On Potentiometry**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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