

# Homework 1: Mixed Ionic/Covalent Compound Naming

*For each of the following questions, determine whether the compound is ionic or covalent and name it appropriately.*

1)  $\text{Na}_2\text{CO}_3$

\_\_\_\_\_

2)  $\text{P}_2\text{O}_5$

\_\_\_\_\_

3)  $\text{NH}_3$

\_\_\_\_\_

4)  $\text{FeSO}_4$

\_\_\_\_\_

5)  $\text{SiO}_2$

\_\_\_\_\_

6)  $\text{GaCl}_3$

\_\_\_\_\_

7)  $\text{CoBr}_2$

\_\_\_\_\_

8)  $\text{B}_2\text{H}_4$

\_\_\_\_\_

9)  $\text{CO}$

\_\_\_\_\_

10)  $\text{P}_4$

\_\_\_\_\_

*For each of the following questions, determine whether the compound is ionic or covalent and write the appropriate formula for it.*

11) dinitrogen trioxide

\_\_\_\_\_

12) nitrogen

\_\_\_\_\_

13) methane

\_\_\_\_\_

14) lithium acetate

\_\_\_\_\_

15) phosphorus trifluoride

\_\_\_\_\_

16) vanadium (V) oxide

\_\_\_\_\_

17) aluminum hydroxide

\_\_\_\_\_

18) zinc sulfide

\_\_\_\_\_

19) silicon tetrafluoride

\_\_\_\_\_

20) silver phosphate

\_\_\_\_\_

# Mixed Ionic Compound Homework

**JR Anderson**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

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