

STOICHIOMETRY: MASS-MASS PROBLEMS

Name _____



How many grams of potassium chloride are produced if 25 g of potassium chlorate decompose?

$$25\text{g KClO}_3 \left(\frac{1\text{ mol}}{122.5\text{g}} \right) \left(\frac{2\text{ mol KCl}}{2\text{ mol KClO}_3} \right) \left(\frac{74.5\text{g KCl}}{1\text{ mol KCl}} \right) = 13.2\text{g KCl}$$



How many grams of hydrogen are necessary to react completely with 50.0 g of nitrogen in the above reaction?

$$50\text{g N}_2 \left(\frac{1\text{ mol N}_2}{28\text{g N}_2} \right) \left(\frac{3\text{ mol H}_2}{1\text{ mol N}_2} \right) \left(\frac{2\text{g H}_2}{1\text{ mol H}_2} \right) = 10.7\text{g H}_2$$

3. How many grams of ammonia are produced in the reaction in Problem 2?

$$50\text{g N}_2 \quad 50\text{g} + 10.7 = \boxed{60.7\text{g NH}_3}$$



How many grams of silver chloride are produced from 5.0 g of silver nitrate reacting with an excess of barium chloride?

$$5.0\text{g AgNO}_3 \left(\frac{1\text{ mol}}{170\text{g}} \right) \left(\frac{1\text{ mol AgCl}}{2\text{ mol AgNO}_3} \right) \left(\frac{143\text{g}}{1\text{ mol AgCl}} \right) = 4.21\text{g AgCl}$$

5. How much barium chloride is necessary to react with the silver nitrate in Problem 4?

$$5.0\text{g AgNO}_3 \left(\frac{1\text{ mol}}{170\text{g AgNO}_3} \right) \left(\frac{1\text{ mol BaCl}_2}{2\text{ mol AgNO}_3} \right) \left(\frac{207\text{g}}{1\text{ mol BaCl}_2} \right) = \boxed{5.0\text{g BaCl}_2}$$

9-7

Mixed Mole Problems Worksheet Answers Instructional Fair

Peipei Pang



Mixed Mole Problems Worksheet Answers Instructional Fair:

Unveiling the Magic of Words: A Report on "**Mixed Mole Problems Worksheet Answers Instructional Fair**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Mixed Mole Problems Worksheet Answers Instructional Fair**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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