Deather

 An old (pre-1987) penny is nearly pure copper. If such a penny has a mass of 3.3 g, how many moles of copper atoms would be in one penny?

 $3.3 \text{ g} \times \underline{1 \text{ mole}}_{63.5 \text{ g}} = 5.2 \times 10^{4} \text{ mol or } 0.052 \text{ mol}$ 

Four nails have a total mass of 4.42 grams. How many moles of iron atoms do they contain?

4.42 g = 1 mole = 7.92 x 10° mol or 0.0792 mol 55.8 g

A raindrop has a mass of 0.050 g. How many moles of water does a raindrop contain?

0.050 g 1 mole = 2.78 x 10° mol → 2.8 x 10° mol

4. What mass of water would you need to have 15.0 moles of H2O?

15.0 moles 18.0 g = 2.70 x 10<sup>2</sup> g

 One box of Morton's Salt contains 737 grams. How many moles of sodium chloride is this?

737 g 1 mole = 12.6 mol

 A chocolate chip cookie recipe calls for 0.050 moles of baking soda (sodium bicarbonate). How many grams should the chef mass out?

0.050 moles 84.0 g = 4.2 g

7. Rust is iron(III) oxide. The owner of a 1959 Cadillac convertible wants to restore it by removing the rust with oxalic acid, but he needs to know how many moles of rust will be involved in the reaction. How many moles of iron(III) oxide are contained in 2 50 kg of rust?

Fe 111.6

48.0

2.50 kg 1000 q 1 mole = 15.7 mol Fe<sub>2</sub>O<sub>3</sub> 1kg 159.6 g CI 35.5

58.5 a

No. 23.0

12.0

48.0

84.0

# **Modeling Chemistry U5 Wsv2 Answers**

**KJ Lindholm-Leary** 

**Modeling Chemistry U5 Wsv2 Answers:** 

Unveiling the Power of Verbal Art: An Psychological Sojourn through Modeling Chemistry U5 Wsv2 Answers

In a world inundated with monitors and the cacophony of instantaneous connection, the profound energy and mental resonance of verbal beauty frequently fade into obscurity, eclipsed by the regular onslaught of sound and distractions. Yet, located within the musical pages of **Modeling Chemistry U5 Wsv2 Answers**, a interesting function of literary splendor that pulses with fresh thoughts, lies an unforgettable journey waiting to be embarked upon. Published by a virtuoso wordsmith, this interesting opus courses viewers on an emotional odyssey, lightly exposing the latent potential and profound affect embedded within the delicate internet of language. Within the heart-wrenching expanse with this evocative evaluation, we will embark upon an introspective exploration of the book is main subjects, dissect its fascinating writing style, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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