Argument Paper, MLA Style (Hammond)

Hammond 1

Jamal Hammond Professor Paschal English 102 17 March XXXX

competition.

Performance Enhancement through Biotechnology

Has No Place in Sports

The debate over athletes' use of performance-enhancing substances is getting more complicated as biotechnologies such as gene therapy become a reality. The availability of these new methods of boosting performance will force us to decide what we value most in sports—displays of physical excellence developed through hard work or victory at all costs. For centuries, spectators and athletes have cherished the tradition of fairness in sports. While sports competition is, of course, largely about winning, it is also about the means by which a player or team wins. Athletes who use any type of biotechnology give themselves an unfair advantage

and disrupt the sense of fair play, and they should be banned from

Researchers are experimenting with techniques that could manipulate an athlete's genetic code to build stronger muscles or increase endurance. Searching for cures for diseases like Parkinson's and muscular dystrophy, scientists at the University of Pennsylvania have created "Schwarzenegger mice," rodents that grew larger-than-normal muscles after receiving injections with a gene that stimulates growth protein. The researchers also found that a combination of gene manipulation and exercise led to a 35% increase in the strength of rats' leg muscles (Lamb 13).

Opening sentences provide background for Hammond's thesis.

Thesis states the main point.

Intermond establishes his credibility by summarising modical research.

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standard hydrochloric acid solution

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web in today s experiment you will determine the solubility product ksp of calcium hydroxide ca oh 2 by measuring the concentration of ca oh 2 in a saturated solution calcium hydroxide is a sparingly soluble salt that dissolves according to the following reaction the solubility product expression for this reaction is

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web use the concentration of calcium hydroxide to calculate the concentration of calcium ions and hydroxide ions in the solution ca oh 2 à ca 2 2oh 5 x 10 2 m calcium ions 5 x 10 2 m 2 0 m hydroxide ions calculate calcium hydroxide ksp using the concentrations of calcium ions and hydroxide ions

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web the chemical equation used was ca oh 2 s ca2 aq 2oh aq the solubility product was determined to be ksp ca2 oh from the titration which was made by dripping a concentrated solution of calcium hydroxide into hydrochloric acid the calcium hydroxide was filtered before starting the titration

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web jul 12 2023 because the concentration of a pure solid such as ca 3 po 4 2 is a constant it does not appear explicitly in the equilibrium constant expression the equilibrium constant expression for the dissolution of calcium phosphate is therefore k ca 2 3 po 3 4 2 ca 3 po 4 2 ca 3 po 4 2 k ksp ca 2 3 po 3 4 2

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