

## Chemistry Unit 8 Worksheet 4

### Samples of Every Kind of Problem

On a separate sheet of paper, write a complete solution to each of the problems below. Follow the procedure outlined in class. Be sure to circle your final answer.

- Calculate the number of moles of potassium chlorate,  $\text{KClO}_3$  (s), that must decompose to produce potassium chloride,  $\text{KCl}$  (s), and 1.8 moles of oxygen gas.

Equation:	$2 \text{KClO}_3 \rightarrow 2 \text{KCl} + 3 \text{O}_2$			1.2 moles $\text{KClO}_3$ needed
Before:	1.2	0	0	
Change:	-1.2	+1.2	+1.8	
After:	0	1.2	1.8	

- In a single displacement reaction, magnesium metal reacts with hydrochloric acid to produce magnesium chloride and hydrogen gas. How many moles of hydrochloric acid are needed to completely react with 2.43 g of magnesium?

Equation:	$\text{Mg} + 2 \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$				0.200 moles $\text{HCl}$ needed
Before:	0.100	?	0	0	
Change:	-0.100	-0.200 +	0.100 +	0.100	
After:	0	?	0.100	0.100	

- Ethane,  $\text{C}_2\text{H}_6$  reacts with oxygen gas to produce carbon dioxide gas and water vapor. What mass of oxygen gas is required to react with 2.20 moles of ethane?

Equation:	$2 \text{C}_2\text{H}_6 + 7 \text{O}_2 \rightarrow 4 \text{CO}_2 + 6 \text{H}_2\text{O}$			
Before:	2.20	xs	0	0
Change:	-2.20	-7.70	+4.40	+6.60
After:	0	xs	4.40	6.60

$$\text{O}_2 \text{ needed: } 7.70 \text{ mole} \times \frac{32.0 \text{ g}}{1 \text{ mole}} = 246 \text{ g O}_2 \text{ required}$$

- Determine the mass of sodium nitrate produced when 0.73 g of nickel (II) nitrate reacts with sodium hydroxide according to the following equation:



$$0.73 \text{ g} \times \frac{1 \text{ mole}}{183 \text{ g}} = .0040 \text{ mole}$$

Equation:	$\text{Ni}(\text{NO}_3)_2 + 2 \text{NaOH} \rightarrow \text{Ni}(\text{OH})_2 + 2 \text{NaNO}_3$			
Before:	0.0040	XS	0	0
Change:	-0.0040	-0.0080	+0.0040	+0.0080
After:	0	XS	0.0040	0.0080

$$\text{NaNO}_3 \text{ produced: } 0.0080 \text{ mole} \times \frac{85.0 \text{ g}}{1 \text{ mole}} = 0.68 \text{ g}$$

# Modeling Chemistry Unit 8 Test

**Yicheng Fang**



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web to help kids break apart multisyllabic words i teach a strategy called spot and dot we dot all of the vowel patterns we see then we break apart the word using those patterns as our guide first place a dot over each vowel sound and connect the first two dots then look at the letters between the dots if there is one consonant draw a line

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web lesson plan decoding subject ela reading grade 3 lesson objective to decode multisyllable words common core standard ccss ela literacy rf 3 3 c decode multisyllable words materials printable student worksheet starter say what do you do when you come to a long hard word when you are reading allow the students to

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web nov 7 2021 while some students will successfully learn to decode multisyllabic words many other kids will stumble over them for years they ll skip the ending or part of the middle or mispronounce the vowel sounds but with our help students can master multisyllabic words and it only takes a handful of minutes each week

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web jan 31 2023 reason 3 gives students a strategy to decode longer words using their known phonics skills teaching students the syllable types will help them access longer words and apply the phonics skills they learned in the primary grades to those words many students skip words or mumble when presented with larger words

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web jul 26 2023 many first grade words also end with the unaccented l sound found in words like little puzzle this is a typical strategy you may use as students begin to decode multisyllabic words explicit instruction to support your students understanding of decoding multisyllabic words you should start by explicitly teaching this concept

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