

## Chemistry – Unit 5 Worksheet 2

1. An old (pre-1987) penny is nearly pure copper. If such a penny has a mass of 3.3 g, how many moles of copper atoms would be in one penny?

$$3.3 \text{ g} \times \frac{1 \text{ mole}}{63.5 \text{ g}} = 5.2 \times 10^{-2} \text{ mol or } 0.052 \text{ mol}$$

2. Four nails have a total mass of 4.42 grams. How many moles of iron atoms do they contain?

$$4.42 \text{ g} \times \frac{1 \text{ mole}}{55.8 \text{ g}} = 7.92 \times 10^{-2} \text{ mol or } 0.0792 \text{ mol}$$

3. A raindrop has a mass of 0.050 g. How many moles of water does a raindrop contain?

$$0.050 \text{ g} \times \frac{1 \text{ mole}}{18.0 \text{ g}} = 2.78 \times 10^{-3} \text{ mol} \rightarrow 2.8 \times 10^{-3} \text{ mol}$$

4. What mass of water would you need to have 15.0 moles of H<sub>2</sub>O?

$$15.0 \text{ moles} \times \frac{18.0 \text{ g}}{1 \text{ mole}} = 2.70 \times 10^2 \text{ g}$$

5. One box of Morton's Salt contains 737 grams. How many moles of sodium chloride is this?

$$737 \text{ g} \times \frac{1 \text{ mole}}{58.5 \text{ g}} = 12.6 \text{ mol}$$

Na	23.0
Cl	35.5
	<hr/> 58.5 g

6. A chocolate chip cookie recipe calls for 0.050 moles of baking soda (sodium bicarbonate). How many grams should the chef mass out?

$$0.050 \text{ moles} \times \frac{84.0 \text{ g}}{1 \text{ mole}} = 4.2 \text{ g}$$

Na	23.0
H	1.0
C	12.0
O	48.0
	<hr/> 84.0

7. Rust is iron(III) oxide. The owner of a 1959 Cadillac convertible wants to restore it by removing the rust with oxalic acid, but he needs to know how many moles of rust will be involved in the reaction. How many moles of iron(III) oxide are contained in 2.50 kg of rust?

$$2.50 \text{ kg} \times \frac{1000 \text{ g}}{1 \text{ kg}} \times \frac{1 \text{ mole}}{159.6 \text{ g}} = 15.7 \text{ mol Fe}_2\text{O}_3$$

Fe	111.6
O	48.0
	<hr/> 159.6

# Modeling Chemistry Unit 5 V2

**AN Whitehead**



## Modeling Chemistry Unit 5 V2:

*Mathematics II | AICTE Prescribed Textbook - English* Garima Singh, 2021-11-01 Mathematics II is a Compulsory paper for the first year students of Diploma engineering courses common to all branches Syllabus of this book is strictly aligned to the model curriculum of AICTE And academic content is amalgamated with the concept of outcome based education Apart from diploma it is useful for all students who are interested in basic elementary mathematics and competitive examinations Book covers seven topics Determinants Matrices Integral Calculus and it s applications Co ordinate Geometry and it s applications vectors and it s applications Differential equations Basic of MATLAB Each topic is written in an easy and lucid manner with a holistic view There has been deliberated attempt to keep the number of pages in the book minimum without compromising with the matter Every chapter contains a set of exercises at the end of each unit to test the student s comprehension Some salient features of the book l For direct recapitulation of main concepts formulae and results a brief summary of each unit has been given l Objective questions and subjective questions are given for practice of students after every unit l Content of the book is aligned with the mapping of Course Outcomes Programs Outcomes and uni Outcomes l Apart from the theory explanation and solved examples book provides for mini projects activities fun facts QR codes case studies video resources etc l The text has been supplemented with notes remarks remember sections within grey boxes l Student and teacher centric subject materials are included in the book in a balanced manner l Real life applications are inserted to improve clarity of this topics l Know more section has been introduced which constitutes of additional information related to the topic l Check out section has been introduced so as to active the curiosity part of the student by correlating all the topics studied in this book with MATLAB l At the end of each unit An excerpt related to eminent Indian Mathematicians is given so as to make l Student have a glimpse of the rich Indian heritage especially in the field of mathematics

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The book delves into Modeling Chemistry Unit 5 V2. Modeling Chemistry Unit 5 V2 is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Modeling Chemistry Unit 5 V2, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Modeling Chemistry Unit 5 V2
- Chapter 2: Essential Elements of Modeling Chemistry Unit 5 V2
- Chapter 3: Modeling Chemistry Unit 5 V2 in Everyday Life
- Chapter 4: Modeling Chemistry Unit 5 V2 in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Modeling Chemistry Unit 5 V2. This chapter will explore what Modeling Chemistry Unit 5 V2 is, why Modeling Chemistry Unit 5 V2 is vital, and how to effectively learn about Modeling Chemistry Unit 5 V2.

3. In chapter 2, the author will delve into the foundational concepts of Modeling Chemistry Unit 5 V2. The second chapter will elucidate the essential principles that must be understood to grasp Modeling Chemistry Unit 5 V2 in its entirety.

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6. In chapter 5, this book will draw a conclusion about Modeling Chemistry Unit 5 V2. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Modeling Chemistry Unit 5 V2.

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