

## Chemistry – Unit 5 Worksheet 2

1. An old (pre-1987) penny is nearly pure copper. If such a penny has a mass of 3.3 g, how many moles of copper atoms would be in one penny?

$$3.3 \text{ g} \times \frac{1 \text{ mole}}{63.5 \text{ g}} = 5.2 \times 10^{-2} \text{ mol or } 0.052 \text{ mol}$$

2. Four nails have a total mass of 4.42 grams. How many moles of iron atoms do they contain?

$$4.42 \text{ g} \times \frac{1 \text{ mole}}{55.8 \text{ g}} = 7.92 \times 10^{-2} \text{ mol or } 0.0792 \text{ mol}$$

3. A raindrop has a mass of 0.050 g. How many moles of water does a raindrop contain?

$$0.050 \text{ g} \times \frac{1 \text{ mole}}{18.0 \text{ g}} = 2.78 \times 10^{-3} \text{ mol} \rightarrow 2.8 \times 10^{-3} \text{ mol}$$

4. What mass of water would you need to have 15.0 moles of H<sub>2</sub>O?

$$15.0 \text{ moles} \times \frac{18.0 \text{ g}}{1 \text{ mole}} = 2.70 \times 10^2 \text{ g}$$

5. One box of Morton's Salt contains 737 grams. How many moles of sodium chloride is this?

$$737 \text{ g} \times \frac{1 \text{ mole}}{58.5 \text{ g}} = 12.6 \text{ mol}$$

|    |              |
|----|--------------|
| Na | 23.0         |
| Cl | 35.5         |
|    | <hr/> 58.5 g |

6. A chocolate chip cookie recipe calls for 0.050 moles of baking soda (sodium bicarbonate). How many grams should the chef mass out?

$$0.050 \text{ moles} \times \frac{84.0 \text{ g}}{1 \text{ mole}} = 4.2 \text{ g}$$

|    |            |
|----|------------|
| Na | 23.0       |
| H  | 1.0        |
| C  | 12.0       |
| O  | 48.0       |
|    | <hr/> 84.0 |

7. Rust is iron(III) oxide. The owner of a 1959 Cadillac convertible wants to restore it by removing the rust with oxalic acid, but he needs to know how many moles of rust will be involved in the reaction. How many moles of iron(III) oxide are contained in 2.50 kg of rust?

$$2.50 \text{ kg} \times \frac{1000 \text{ g}}{1 \text{ kg}} \times \frac{1 \text{ mole}}{159.6 \text{ g}} = 15.7 \text{ mol Fe}_2\text{O}_3$$

|    |             |
|----|-------------|
| Fe | 111.6       |
| O  | 48.0        |
|    | <hr/> 159.6 |

# Modeling Chemistry Unit Answers

**Agnes French Jaques**



## **Modeling Chemistry Unit Answers:**

*Robotics Interview Questions and Answers* Manish Soni, 2024-11-13 Welcome to Robotics Interview Questions and Answers a comprehensive guide designed to navigate the dynamic world of robotics through a lens of inquiry and exploration In the pages that follow you will embark on a journey through the fascinating realm of robotics uncovering a myriad of topics that span the breadth and depth of this transformative field This book seeks to provide not only a wealth of knowledge but also a practical resource for individuals aspiring to delve into the world of robotics or those seeking to enhance their understanding of its myriad facets In the age of automation artificial intelligence and the Internet of Things robotics has emerged as a pivotal force shaping our future From manufacturing floors to healthcare settings from deep space exploration to our own living rooms robots have become an integral part of our daily lives Whether you seek to gain knowledge for interviews academic pursuits or simply to satisfy your curiosity about the incredible world of robotics this book is designed to be your trusted companion It serves as a roadmap to understanding the fundamentals the nuances and the future possibilities that robotics holds

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**Chemistry in the Community (Enhanced Core Four)** American Chemical Society, 2006-02-15

*Learning Progressions in Science* Alicia C. Alonzo, Amelia Wenk Gotwals, 2012-07-30 Learning progressions descriptions of increasingly sophisticated ways of thinking about or understanding a topic National Research Council 2007 represent a promising framework for developing organized curricula and meaningful assessments in science In addition well grounded learning progressions may allow for coherence between cognitive models of how understanding develops in a given domain classroom instruction professional development and classroom and large scale assessments Because of the promise that learning progressions hold for bringing organization and structure to often disconnected views of how to teach and assess science they are rapidly gaining popularity in the science education community However there are significant challenges faced by all engaged in this work In June 2009 science education researchers and practitioners as well as scientists psychometricians and assessment specialists convened to discuss these challenges as part of the Learning Progressions in Science LeaPS conference The LeaPS conference provided a structured forum for considering design decisions entailed in four aspects of work on learning progressions defining learning progressions developing assessments to elicit student responses relative to learning progressions modeling and interpreting student performance with respect to a learning progression and using learning progressions to influence standards curricula and teacher education This book presents specific examples of learning progression work and syntheses of ideas from these examples and discussions at the LeaPS

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**Quarterly Journal of the Florida Academy of Sciences** Florida Academy of Sciences,1993     **Resources in Education** ,1982     *Chemical Process Analysis and Design Using Computers* Working Party on the Use of Computers in Chemical Engineering. Symposium,1982     *The Application of Mathematical Modelling to Process Development and Design* L. M. Rose,1974     *Modelling the Fate of Chemicals in the Environment* Ian D. Moore,1991     **Chemical Engineering Progress** ,2007     **The School World** ,1908     **Modelling and Computation for Applications in Mathematics, Science, and Engineering** Joseph W. Jerome,1998 This book discusses many of the connections between cutting edge applications and advanced numerical and algorithmic techniques Topics include models for long range molecular forces biological and chemical contamination including hierarchical problem solving ideas from computer science discrete mathematics including connections to the buckyball structure of carbon 60 dimension reduction techniques in incompressible fluid mechanics aspects of charge transport bridging compressible fluids gas dynamics and semiconductors the approximation problem in control theory questions related to weighted approximation by polynomials in the complex plane and high friction limits of hydrodynamic models     *The Journal of Proceedings and Addresses of the National Educational Association* National Educational Association (U.S.),1887     **Proceedings of the Hydrologic Transport Modeling Symposium, December 10-11, 1979, Hyatt-Regency Hotel, New Orleans, Louisiana** American Society of Agricultural Engineers,1980

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