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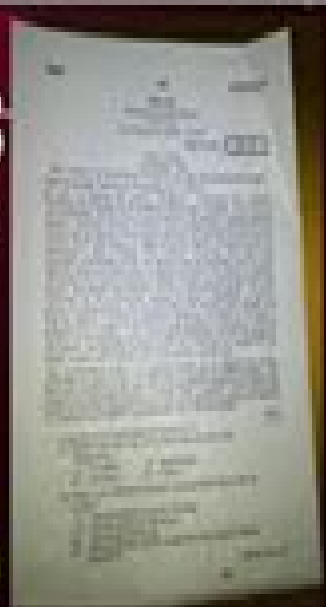
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English Interrogative Sentences: Common Interrogative Patterns Manik Joshi, 2014-10-25 This Book Covers The Following Topics What are Interrogative Sentences Structure 1 Wh Question Word Be Do Have Modal 1A What Be Do Have Modal 1B When Be Do Have Modal 1C Where Be Do Have Modal 1D Which Be Do Have Modal 1E Who Be Do Have Modal 1F Whom Be Do Have Modal 1G Whose Be Do Have Modal 1H Why Be Do Have Modal 1I How Be Do Have Modal Structure 2 Wh Question Word Word Words Be Do Have Modal 2A What Word Words Be Do Have Modal 2B When Word Words Be Do Have Modal 2C Where Word Words Be Do Have Modal 2D Which Word Words Be Do Have Modal 2E Who Word Words Be Do Have Modal 2F Whom Word Words Be Do Have Modal 2G Whose Word Words Be Do Have Modal 2H Why Word Words Be Do Have Modal 2I How Word Words Be Do Have Modal Structure 3 Wh Question Word Main Verb Present or Past Structure 4 Interrogatives Sentences Be Do Have Modal 4A Interrogatives Starting From Am Is Are Was Were 4B Interrogatives Starting From Do Does Did 4C Interrogatives Starting From Have Has Had 4D Interrogatives Starting From Modal Verbs Structure 5 Question Tags Structure 6 What if Structure 7 How Long How Much How Many Structure 8 Wh Question Word To Verb Word Structure 9 What About and How About Structure 10 Alternative Questions Structure 11 Indirect Questions Formation of Interrogatives from Affirmatives Exercises Sample This What are Interrogative Sentences Interrogative sentences are used to ask questions An interrogative sentence ends with a question mark The most common interrogative words are as follows What When Where Which Who Whom Whose Why How Interrogative words and what they refer to What refers to specific information or confirmation repetition When refers to at what time or on what occasion Where refers to in what place position or situation Which refers to choice or alternative Who refers to identity of a subject person people Whom refers to identity of an object person people Whose refers to who something belongs to Why refers to reason explanation or purpose How refers to way or manner condition or quality These words are called Wh question words because all these words contain the letter w and h All these words except how even start from Wh NOTE The following words are also used to ask questions Whatever Whenever Wherever Whoever These forms show surprise confusion or emphasis Besides Wh question words Auxiliary Verbs Be Do Have and Modal Verbs are also used to form interrogative sentences Following is the list of auxiliary and modal verbs Auxiliary Verb Be Am Is Are Was Were Auxiliary Verb Do Do Does Did Auxiliary Verb Have Have Has Had Modal Verbs May Might Can Could Will Would Shall Should Must Need Used To Ought To Dare You can begin sentences with these verbs to form Yes No interrogative sentences 1A What Be Do Have Modal What is a good pet to give a five year old child What is a long way away What is a reasonable grocery budget What is age got to do with it What is all that What is Australia s national food What is behind the nation s food shortages What is better for your company happy staff or short term profits What is Brazil to you What is going on in India What is going to take place over the next 90 minutes What is in the haze we are breathing What is it about the first day of the year that gets us so excited What is it and does it work What is

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web aug 19 2022 atoms with the same number of protons but different numbers of neutrons are called isotopes they share almost the same chemical properties but differ in mass and therefore in physical properties there are stable isotopes which do not emit radiation and there are unstable isotopes which do emit radiation

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nuclei contain where neutrons are massive particles of zero charge a nucleus may contain variable numbers of neutrons and this gives rise to the phenomenon of isotopes

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web verified by toppr correct option is d atoms having same atomic number but different atomic mass are known as isotopes the difference in the mass is due to the difference in their number of neutrons with the same number of electrons and protons

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web part i answer the questions based on the above reading what is an isotope isotopes are versions of the same element they have the same number of protons and electrons as the element but different mass numbers and number of neutrons what does the number next to isotopes signify the number indicates the isotope s mass number

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