



सामान्य निर्देश :

निम्नलिखित निर्देशों को बहुत सावधानी से पढ़िए और उनका साख्ती से पालन कीजिए :

- (i) यह प्रश्न-पत्र चार खण्डों में विभाजित किया गया है – क, ख, ग और घ ।
- (ii) इस प्रश्न-पत्र में 37 प्रश्न हैं । सभी प्रश्न अनिवार्य हैं ।
- (iii) खण्ड क – प्रश्न संख्या 1 से 20 तक अति लघु-उत्तरीय प्रश्न हैं, प्रत्येक प्रश्न 1 अंक का है ।
- (iv) खण्ड ख – प्रश्न संख्या 21 से 27 तक लघु-उत्तरीय प्रश्न हैं, प्रत्येक प्रश्न 2 अंकों का है ।
- (v) खण्ड ग – प्रश्न संख्या 28 से 34 तक दीर्घ-उत्तरीय प्रकार के प्रश्न हैं, प्रत्येक प्रश्न 3 अंकों का है ।
- (vi) खण्ड घ – प्रश्न संख्या 35 से 37 तक भी दीर्घ-उत्तरीय प्रकार के प्रश्न हैं, प्रत्येक प्रश्न 5 अंकों का है ।
- (vii) प्रश्न-पत्र में कोई समग्र विकल्प नहीं है । तथापि, एक-एक अंक के दो प्रश्नों में, दो-दो अंकों वाले दो प्रश्नों में, तीन-तीन अंकों वाले एक प्रश्न में तथा पाँच-पाँच अंकों वाले तीनों प्रश्नों में आंतरिक विकल्प दिया गया है । ऐसे प्रश्नों में केवल एक ही विकल्प का उत्तर दीजिए ।
- (viii) इसके अतिरिक्त, आवश्यकतानुसार, प्रत्येक खण्ड और प्रश्न के साथ यथोचित निर्देश दिए गए हैं ।
- (ix) कैल्कुलेटोर्स अथवा लॉग टेबलों के प्रयोग की अनुमति नहीं है ।
- (x) जहाँ आवश्यक हो, आप निम्नलिखित भौतिक नियतांकों के मानों का उपयोग कर सकते हैं :

$$c = 3 \times 10^8 \text{ m/s}$$

$$h = 6.63 \times 10^{-34} \text{ Js}$$

$$e = 1.6 \times 10^{-19} \text{ C}$$

$$\mu_0 = 4\pi \times 10^{-7} \text{ T m A}^{-1}$$

$$\epsilon_0 = 8.854 \times 10^{-12} \text{ C}^2 \text{ N}^{-1} \text{ m}^{-2}$$

$$\frac{1}{4\pi\epsilon_0} = 9 \times 10^9 \text{ N m}^2 \text{ C}^{-2}$$

$$\text{इलेक्ट्रॉन का द्रव्यमान (m}_e\text{)} = 9.1 \times 10^{-31} \text{ kg}$$

$$\text{न्यूट्रॉन का द्रव्यमान} = 1.675 \times 10^{-27} \text{ kg}$$

$$\text{प्रोटॉन का द्रव्यमान} = 1.673 \times 10^{-27} \text{ kg}$$

$$\text{आवोगाद्रो संख्या} = 6.023 \times 10^{23} \text{ प्रति ग्राम मोल}$$

$$\text{बोल्ट्ज़मान नियतांक} = 1.38 \times 10^{-23} \text{ JK}^{-1}$$

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Table of Contents Jsce Examination In All Papers 24

1. Understanding the eBook Jsce Examination In All Papers 24
 - The Rise of Digital Reading Jsce Examination In All Papers 24
 - Advantages of eBooks Over Traditional Books
2. Identifying Jsce Examination In All Papers 24
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Jsce Examination In All Papers 24
 - User-Friendly Interface
4. Exploring eBook Recommendations from Jsce Examination In All Papers 24
 - Personalized Recommendations
 - Jsce Examination In All Papers 24 User Reviews and Ratings
 - Jsce Examination In All Papers 24 and Bestseller Lists

5. Accessing Jsce Examination In All Papers 24 Free and Paid eBooks
 - Jsce Examination In All Papers 24 Public Domain eBooks
 - Jsce Examination In All Papers 24 eBook Subscription Services
 - Jsce Examination In All Papers 24 Budget-Friendly Options
6. Navigating Jsce Examination In All Papers 24 eBook Formats
 - ePub, PDF, MOBI, and More
 - Jsce Examination In All Papers 24 Compatibility with Devices
 - Jsce Examination In All Papers 24 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Jsce Examination In All Papers 24
 - Highlighting and Note-Taking Jsce Examination In All Papers 24
 - Interactive Elements Jsce Examination In All Papers 24
8. Staying Engaged with Jsce Examination In All Papers 24
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Jsce Examination In All Papers 24
9. Balancing eBooks and Physical Books Jsce Examination In All Papers 24
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Jsce Examination In All Papers 24
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Jsce Examination In All Papers 24
 - Setting Reading Goals Jsce Examination In All Papers 24
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Jsce Examination In All Papers 24
 - Fact-Checking eBook Content of Jsce Examination In All Papers 24
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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