

- (b) Lithium, carbon (in the form of diamond) and fluorine have very different melting points.

These differences in melting points are the result of different types of structure and different forces or bonds between the particles in the structures.

Part of the table below has been filled in.

Complete the table below.

	<b>Lithium</b>	<b>Carbon (diamond)</b>	<b>Fluorine</b>
<b>Melting point/°C</b>	181	3550	–220
<b>Structure</b>	Giant		Simple
<b>Force or bond overcome on melting</b>	Metallic bond		
<b>Particles between which the force or bond is acting</b>			

[6]

[Total: 10]

**END OF QUESTION PAPER**

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**Daniela Niemeyer**

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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## **Table of Contents Jun 2013 Ocr F58paper**

1. Understanding the eBook Jun 2013 Ocr F58paper
  - The Rise of Digital Reading Jun 2013 Ocr F58paper
  - Advantages of eBooks Over Traditional Books
2. Identifying Jun 2013 Ocr F58paper
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Jun 2013 Ocr F58paper
  - User-Friendly Interface
4. Exploring eBook Recommendations from Jun 2013 Ocr F58paper
  - Personalized Recommendations
  - Jun 2013 Ocr F58paper User Reviews and Ratings
  - Jun 2013 Ocr F58paper and Bestseller Lists
5. Accessing Jun 2013 Ocr F58paper Free and Paid eBooks
  - Jun 2013 Ocr F58paper Public Domain eBooks
  - Jun 2013 Ocr F58paper eBook Subscription Services
  - Jun 2013 Ocr F58paper Budget-Friendly Options
6. Navigating Jun 2013 Ocr F58paper eBook Formats
  - ePub, PDF, MOBI, and More
  - Jun 2013 Ocr F58paper Compatibility with Devices
  - Jun 2013 Ocr F58paper Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Jun 2013 Ocr F58paper
  - Highlighting and Note-Taking Jun 2013 Ocr F58paper
  - Interactive Elements Jun 2013 Ocr F58paper
8. Staying Engaged with Jun 2013 Ocr F58paper

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Jun 2013 Ocr F58paper
- 9. Balancing eBooks and Physical Books Jun 2013 Ocr F58paper
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Jun 2013 Ocr F58paper
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Jun 2013 Ocr F58paper
  - Setting Reading Goals Jun 2013 Ocr F58paper
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Jun 2013 Ocr F58paper
  - Fact-Checking eBook Content of Jun 2013 Ocr F58paper
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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