Analytical Chemistry MCQ Test. September'21.

- The metal hydroxide which is insoluble in caustic soda but soluble in excess of ammonium hydroxide.
- ammonium hydroxide.

 1. Cu(OH): 2 Zn(OH): 3 Fe(OH): 4 None
- 2. The salt solution which does not react with ammonium hydroxide is ------
- 1. Calcium nitrate 2. Zinc nitrate
- 3.Lead nitrate 4. Copper nitrate
- 3. The metal oxide which can react with acids as well as alkali is------
 - 1. Silver oxide 2. Copper (II) oxide
 - 3. Aluminum oxide 4. Calcium oxide
 - 4.Al metal shows amphoteric behaviour.it can be shown by
 - 1. Its reaction with dil.NaOH only.
 - Its reaction with dil.HCl only.
 - 3. Its reaction with both dil.HCl and dil.NaOH.
 - It does not react with either dil.HCl or dil.NaOH

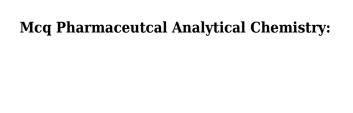
QUESTION 5

Which of the following hydroxides soluble in excess of NaOH?

- 1. Cu (OH)₂ 2. Zn (OH)₂ 3. Fe (OH)₂ 4Fe(OH)₃
- The salt which gives a dirty green precipitate with sodium hydroxide solution is.
 - 1. Iron (III) sulphate 2. Iron (II) sulphate
 - 3.Copper sulphate 4. Zinc sulphate
- The salt which gives a reddish-brown precipitate with sodium hydroxide solution is.
 - 1. Iron (III) sulphate 2. Iron (II) sulphate
 - 3.Copper sulphate 4. Zinc sulphate
- The salt which gives a gelatinous white precipitate with Ammonium hydroxide solution is.
 - 1. Iron (III) choride 2. Iron (II) sulphate
 - 3.Copper sulphate 4. Zinc sulphate
- The salt which gives a chalky white precipitate with Ammonium hydroxide solution is.
 - 1. Lead nitrate 2. Iron (II) sulphate
 - 3.Copper sulphate 4. Zinc sulphate

Mcq Pharmaceutcal Analytical Chemistry

Christian G. Meyer



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- 1. Understanding the eBook Mcq Pharmaceutcal Analytical Chemistry
 - The Rise of Digital Reading Mcq Pharmaceutcal Analytical Chemistry
 - $\circ\,$ Advantages of eBooks Over Traditional Books
- 2. Identifying Mcg Pharmaceutcal Analytical Chemistry
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 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mcq Pharmaceutcal Analytical Chemistry
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Mcq Pharmaceutcal Analytical Chemistry
 - Personalized Recommendations
 - Mcq Pharmaceutcal Analytical Chemistry User Reviews and Ratings
 - Mcq Pharmaceutcal Analytical Chemistry and Bestseller Lists
- 5. Accessing Mcg Pharmaceutcal Analytical Chemistry Free and Paid eBooks
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 - Interactive Elements Mcq Pharmaceutcal Analytical Chemistry
- 8. Staying Engaged with Mcg Pharmaceutcal Analytical Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Mcq Pharmaceutcal Analytical Chemistry
- 9. Balancing eBooks and Physical Books Mcq Pharmaceutcal Analytical Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Mcg Pharmaceutcal Analytical Chemistry
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Mcg Pharmaceutcal Analytical Chemistry
 - Setting Reading Goals Mcq Pharmaceutcal Analytical Chemistry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Mcq Pharmaceutcal Analytical Chemistry
 - Fact-Checking eBook Content of Mcq Pharmaceutcal Analytical Chemistry
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
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- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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