

# JOMO KENYATTA UNIVERSITY OF AGRICULTURE AND TECHNOLOGY

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# Jomo Kenyatta University September 2015

**SA Adler**



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web solution a  $3.8 \times 10^{-51}$  grams b  $4.11 \times 10^{-45}$  grams c  $7.31 \times 10^{-25}$  grams 4 7 1 mole practice with guidance exercises is shared under a not declared license and was authored remixed and or curated by libretexts these are homework exercises to accompany the textmap created for chemistry a molecular approach by nivaldo tro

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web what is the mole fraction of licl in the solution 11 8 practice problem acetone  $\text{C}_3\text{H}_6\text{O}$  has a density of  $0.791 \text{ g/ml}$  and xylene  $\text{C}_8\text{H}_{10}$  has a density of  $0.864 \text{ g/ml}$  a solution was prepared by mixing acetone and xylene calculate the mole fraction of acetone in a solution if  $19.0 \text{ ml}$  of acetone was dissolved in  $98.4 \text{ ml}$  of xylene 7 1

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web since the mole fraction of the  $\text{O}_2$  is  $0.050$  it contributes  $0.050$  or if you will  $5.0$  of the total pressure thus total pressure is  $0.2105042 \text{ atm}$  problem 9 a sample of  $1.43 \text{ g}$  of helium and an unweighed quantity of  $\text{O}_2$  are mixed in a flask at room temperature