

# JEE Main 2020 Paper



**Sol. 2**



P - OH bonds = 4

P = O bonds = 2

P - O - P linkage = 1

Ans. 4, 2, 1

**21.** The mole fraction of glucose ( $\text{C}_6\text{H}_{12}\text{O}_6$ ) in an aqueous binary solution is 0.1. The mass percentage of water in it, to the nearest integer, is \_\_\_\_\_.

**Sol. 47 %**

$$x_{\text{Glucose}} = 0.1$$

$$\begin{aligned} \text{mass\% of glucose} &= \frac{0.1 \times 180}{0.1 \times 180 + 0.9 \times 18} \times 100 \\ &= \frac{1800}{18 + 16.2} \\ &= \frac{1800}{34.2} \\ &= 52.63\% \\ &= 53\% \end{aligned}$$

$$\therefore \text{mass \% of H}_2\text{O} = 47\%$$

**22.** The volume strength of 8.9 M  $\text{H}_2\text{O}_2$  solution calculated at 273 K and 1 atm is \_\_\_\_\_. ( $R = 0.0821 \text{ L atm K}^{-1} \text{ mol}^{-1}$ ) (rounded off to the nearest integer)

**Sol. 100**

$$\begin{aligned} \text{Vol. strength} &= \frac{8.9}{2} \times \frac{0.0821 \times 273}{1} \\ &= 99.73 \\ &= 100 \end{aligned}$$

**23.** An element with molar mass  $2.7 \times 10^{-2} \text{ kg mol}^{-1}$  forms a cubic unit cell with edge length 405 pm. If its density is  $2.7 \times 10^3 \text{ kg m}^{-3}$ , the radius of the element is approximately \_\_\_\_\_  $\times 10^{-12} \text{ m}$  (to the nearest integer).

**Sol. 143**

$$\text{Density} = \frac{Z \times \text{GMM}}{N_A \times a^3}$$

$$\begin{aligned} 2.7 \times 10^3 &= \frac{Z \times 2.7 \times 10^{-2}}{6.023 \times 10^{23} \times (405 \times 10^{-12})^3} \\ Z &= 6.023 \times 405 \times 405 \times 405 \times 10^{23-36+3+2} \\ Z &= 6.023 \times 405 \times 405 \times 405 \times 10^{-8} \\ Z &= 4 \end{aligned}$$

FCC

$$4R = \sqrt{2} \times a$$

$$R = \frac{405}{2\sqrt{2}} \times 10^{-12} = 143.21 \times 10^{-12} \text{ m}$$

= 143 ans

# Jee Mains 2question Paper Solution By Fiit Jeee

**RS Peters**



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**JEE Main 13 Years Question Paper** Prepjee,2014-08-01 THIS BOOK HAS BEEN DEVELOPED AND DESIGNED TO OVERCOME WITH PRESSING NEED OF JEE MAIN ASPIRANTS THE BOOK CONTAINS PAPERS OF AIEEE JEE MAIN OF PAST 13 YEARS WITH ANSWER KEY AND DETAILED SOLUTION FOR THREE YEARS PAPERS THIS WILL GUIDE THE COURSE OF PREPARATION OF AN ASPIRANT IN A DEFINITE AND REQUIRED DIRECTION LIKE RADAR MOREOVER SOLUTION OF GIVEN YEARS i e 2012 2014 IS IN APPROPRIATE MANNER SO THAT AVERAGE STUDENT CAN UNDERSTANT AND SOLVE PROBLEMS EASILY THIS BOOK ALSO HELPS STUDENTS IN CONCEPT BUILDING **16**

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### **Table of Contents Jee Mains 2question Paper Solution By Fiit Jee**

1. Understanding the eBook Jee Mains 2question Paper Solution By Fiit Jee
  - The Rise of Digital Reading Jee Mains 2question Paper Solution By Fiit Jee
  - Advantages of eBooks Over Traditional Books
2. Identifying Jee Mains 2question Paper Solution By Fiit Jee
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an eBook Jee Mains 2question Paper Solution By Fiit Jee
  - User-Friendly Interface
4. Exploring eBook Recommendations from Jee Mains 2question Paper Solution By Fiit Jee
  - Personalized Recommendations

- Jee Mains 2question Paper Solution By Fiit Jeee User Reviews and Ratings
- Jee Mains 2question Paper Solution By Fiit Jeee and Bestseller Lists
- 5. Accessing Jee Mains 2question Paper Solution By Fiit Jeee Free and Paid eBooks
  - Jee Mains 2question Paper Solution By Fiit Jeee Public Domain eBooks
  - Jee Mains 2question Paper Solution By Fiit Jeee eBook Subscription Services
  - Jee Mains 2question Paper Solution By Fiit Jeee Budget-Friendly Options
- 6. Navigating Jee Mains 2question Paper Solution By Fiit Jeee eBook Formats
  - ePub, PDF, MOBI, and More
  - Jee Mains 2question Paper Solution By Fiit Jeee Compatibility with Devices
  - Jee Mains 2question Paper Solution By Fiit Jeee Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Jee Mains 2question Paper Solution By Fiit Jeee
  - Highlighting and Note-Taking Jee Mains 2question Paper Solution By Fiit Jeee
  - Interactive Elements Jee Mains 2question Paper Solution By Fiit Jeee
- 8. Staying Engaged with Jee Mains 2question Paper Solution By Fiit Jeee
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Jee Mains 2question Paper Solution By Fiit Jeee
- 9. Balancing eBooks and Physical Books Jee Mains 2question Paper Solution By Fiit Jeee
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Jee Mains 2question Paper Solution By Fiit Jeee
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Jee Mains 2question Paper Solution By Fiit Jeee
  - Setting Reading Goals Jee Mains 2question Paper Solution By Fiit Jeee
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Jee Mains 2question Paper Solution By Fiit Jeee
  - Fact-Checking eBook Content of Jee Mains 2question Paper Solution By Fiit Jeee

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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