KISA Pre Board Examination - Answer key SUBJECT-CHEMISTRY

Section A

1x15=15M

Ouestion 1.

i.	a), they have similar electronic configuration in valence shell	
ii.	e). CuSO ₄	
iii.	b). Oxygen is released at anode.	
žv.	e). Substitution	
W.	a). He	
wi.	(e). Consists of molecules	
vii.	b). Lead	
riii.	d). Argon	
in.	a). Magnesium oxide	
x.	(b). Aqueous acetic acid	
xi.	c). Reacts to form water, nitrogen dioxide and carbon dioxide	63
xii.	a). K ₂ SO ₄	
ciii.	e).1	
civ.	e). Fe, Cr and Ni	
XV.	a). Decreases	
Ques	on 2.	
(i	(a) A>Cone.sulphuric acid	(3)
	B→ sodium nitrate/potassium nitrate	
	C→ Nitric acid	
	(b)4HNO ₃	(1)
	(e)Cu+4 HNO ₃	(1)
(i	Column A Column B	1x5=5M
	a) Sodium chloride - ionic bond	
	b) Ammonium ion - covalent and co	-ordinate bond
	e) Electronegativity across the period -I increases	
	d) Nonmetallic character down the group - decrease	

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 - The Rise of Digital Reading Jeemainpaper 2 Ans Key 2014 Code K
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Jeemainpaper2 Ans Key 2014 Code K
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Jeemainpaper Ans Key 2014 Code K
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Jeemainpaper2 Ans Key 2014 Code K
 - Personalized Recommendations
 - Jeemainpaper2 Ans Key 2014 Code K User Reviews and Ratings
 - Jeemainpaper2 Ans Key 2014 Code K and Bestseller Lists
- 5. Accessing Jeemainpaper 2 Ans Key 2014 Code K Free and Paid eBooks

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- Jeemainpaper2 Ans Key 2014 Code K Budget-Friendly Options
- 6. Navigating Jeemainpaper 2 Ans Key 2014 Code K eBook Formats
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 - Jeemainpaper2 Ans Key 2014 Code K Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Jeemainpaper2 Ans Key 2014 Code K
 - Highlighting and Note-Taking Jeemainpaper 2Ans Key 2014 Code K
 - Interactive Elements Jeemainpaper 2 Ans Key 2014 Code K
- 8. Staying Engaged with Jeemainpaper2 Ans Key 2014 Code K
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Jeemainpaper2 Ans Key 2014 Code K
- 9. Balancing eBooks and Physical Books Jeemainpaper2 Ans Key 2014 Code K
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Jeemainpaper 2 Ans Key 2014 Code K
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Jeemainpaper 2 Ans Key 2014 Code K
 - Setting Reading Goals Jeemainpaper2 Ans Key 2014 Code K
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Jeemainpaper2 Ans Key 2014 Code K
 - Fact-Checking eBook Content of Jeemainpaper2 Ans Key 2014 Code K
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
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