

Chemistry – Unit 5 Worksheet 2

1. An old (pre-1987) penny is nearly pure copper. If such a penny has a mass of 3.3 g, how many moles of copper atoms would be in one penny?

$$3.3 \text{ g} \times \frac{1 \text{ mole}}{63.5 \text{ g}} = 5.2 \times 10^{-2} \text{ mol or } 0.052 \text{ mol}$$

2. Four nails have a total mass of 4.42 grams. How many moles of iron atoms do they contain?

$$4.42 \text{ g} \times \frac{1 \text{ mole}}{55.8 \text{ g}} = 7.92 \times 10^{-2} \text{ mol or } 0.0792 \text{ mol}$$

3. A raindrop has a mass of 0.050 g. How many moles of water does a raindrop contain?

$$0.050 \text{ g} \times \frac{1 \text{ mole}}{18.0 \text{ g}} = 2.78 \times 10^{-3} \text{ mol} \rightarrow 2.8 \times 10^{-3} \text{ mol}$$

4. What mass of water would you need to have 15.0 moles of H₂O?

$$15.0 \text{ moles} \times \frac{18.0 \text{ g}}{1 \text{ mole}} = 2.70 \times 10^2 \text{ g}$$

5. One box of Morton's Salt contains 737 grams. How many moles of sodium chloride is this?

$$737 \text{ g} \times \frac{1 \text{ mole}}{58.5 \text{ g}} = 12.6 \text{ mol}$$

Na	23.0
Cl	35.5
	58.5 g

6. A chocolate chip cookie recipe calls for 0.050 moles of baking soda (sodium bicarbonate). How many grams should the chef mass out?

$$0.050 \text{ moles} \times \frac{84.0 \text{ g}}{1 \text{ mole}} = 4.2 \text{ g}$$

Na	23.0
H	1.0
C	12.0
O	48.0
	84.0

7. Rust is iron(III) oxide. The owner of a 1959 Cadillac convertible wants to restore it by removing the rust with oxalic acid, but he needs to know how many moles of rust will be involved in the reaction. How many moles of iron(III) oxide are contained in 2.50 kg of rust?

$$2.50 \text{ kg} \times \frac{1000 \text{ g}}{1 \text{ kg}} \times \frac{1 \text{ mole}}{159.6 \text{ g}} = 15.7 \text{ mol Fe}_2\text{O}_3$$

Fe	111.6
O	48.0
	159.6

Modeling Chemistry Answers Key U5 Test B V2

G Thomas



Modeling Chemistry Answers Key U5 Test B V2 :

This book delves into Modeling Chemistry Answers Key U5 Test B V2 . Modeling Chemistry Answers Key U5 Test B V2 is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Modeling Chemistry Answers Key U5 Test B V2 , encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Modeling Chemistry Answers Key U5 Test B V2
- Chapter 2: Essential Elements of Modeling Chemistry Answers Key U5 Test B V2
- Chapter 3: Modeling Chemistry Answers Key U5 Test B V2 in Everyday Life
- Chapter 4: Modeling Chemistry Answers Key U5 Test B V2 in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Modeling Chemistry Answers Key U5 Test B V2 . This chapter will explore what Modeling Chemistry Answers Key U5 Test B V2 is, why Modeling Chemistry Answers Key U5 Test B V2 is vital, and how to effectively learn about Modeling Chemistry Answers Key U5 Test B V2 .
3. In chapter 2, the author will delve into the foundational concepts of Modeling Chemistry Answers Key U5 Test B V2 . The second chapter will elucidate the essential principles that must be understood to grasp Modeling Chemistry Answers Key U5 Test B V2 in its entirety.
4. In chapter 3, this book will examine the practical applications of Modeling Chemistry Answers Key U5 Test B V2 in daily life. The third chapter will showcase real-world examples of how Modeling Chemistry Answers Key U5 Test B V2 can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Modeling Chemistry Answers Key U5 Test B V2 in specific contexts. The fourth chapter will explore how Modeling Chemistry Answers Key U5 Test B V2 is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Modeling Chemistry Answers Key U5 Test B V2 . The final chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Modeling Chemistry Answers Key U5 Test B V2 .

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