

6. When one gram acid of  $\text{KMnO}_4$  reacts with  $\text{HCl}$ , the volume of chlorine liberated at NTP will be
- (a) 11.2 litres (b) 22.4 litres  
(c) 44.8 litres (d) 56.8 litres
7. When  $\text{MnO}_4^-$  ions reacts with  $\text{Br}^-$  ions in acid solution  $\text{Br}_2$  is liberated. The equivalent weight of  $\text{KMnO}_4$  in this reaction is
- (a) 500 (b) 1500  
(c) 300 (d) 1000
8. 54 g. of hydrogen peroxide is present in 1000 ml. of solution. This solution is called
- (a) 10 vol solution (b) 20 vol solution  
(c) 30 vol solution (d) 50 vol solution
9. The number of moles of  $\text{KMnO}_4$  that will be needed to react completely with one mole of ferrous oxalate in acidic solution is
- (a) 10 (b) 20  
(c) 40 (d) 2
10. To prepare a solution that is 0.05 M  $\text{KCl}$  starting with 100 mL of 0.40 M  $\text{KCl}$
- (a) add 0.30 g  $\text{KCl}$   
(b) add 50 mL of water  
(c) add 0.30 mol  $\text{KCl}$   
(d) evaporate 10 mL water
11. In hot alkaline solution,  $\text{Br}_2$  disproportionates to  $\text{Br}^-$  and  $\text{BrO}_2^-$
- $$3\text{Br}_2 + 6\text{OH}^- \rightarrow 5\text{Br}^- + \text{BrO}_2^- + 3\text{H}_2\text{O}$$
- Atomic equivalent weight of  $\text{Br}_2$  is (molecular weight = 160)
- (a)  $\frac{80}{5}$  (b)  $\frac{80}{3}$   
(c)  $\frac{160}{5}$  (d)  $\frac{160}{3}$
12.  $(\text{Fe}^{2+})$  in a solution prepared by adding 20.00 mL of 0.10 M  $\text{FeSO}_4$  with 70 mL of 0.10 M  $\text{Na}_2\text{SO}_4$  is
- (a) 0.020 M (b) 0.140 M  
(c) 0.010 M (d) 0.200 M
13. When 50 mL of 0.100 M  $\text{HCl}$  is mixed with 100 mL of 0.10 M  $\text{KOH}$ , the resultant solution is the same as a solution of
- (a) 0.10 M  $\text{HCl}$  and 0.05 M  $\text{HCl}$   
(b) 0.05 M  $\text{HCl}$   
(c) 0.05 M  $\text{HCl}$  and 0.05 M  $\text{KOH}$   
(d) 0.05 M  $\text{HCl}$  and 0.05 M  $\text{HCl}$
14. 50 mL of  $\text{H}_2\text{O}_2$  solution contains strength = x% required 10 mL of  $\text{KMnO}_4$  solution in acidic medium. Hence x is
- (a) 0.20 (b) 0.1  
(c) 0.1 (d) 0.0
15. Molarity of 1.0 M  $\text{H}_2\text{SO}_4$  (d = 1.8 mL/cm<sup>3</sup>) is
- (a) 36 mol kg<sup>-1</sup> (b) 300 mol kg<sup>-1</sup>

# Multiple Choice Questions On Inorganic Chemistry

**Thomas Ward (F.C.S.)**



## **Multiple Choice Questions On Inorganic Chemistry:**

## Reviewing **Multiple Choice Questions On Inorganic Chemistry**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Multiple Choice Questions On Inorganic Chemistry**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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saw pulley weight motor pulley and drive belt are separately wrapped and packed under the saw base remove rust preventive from machined surfaces with kerosene

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web apr 10 2021 the machine and machine tool does consist of different parts and mechanism for its working as the power hacksaw is an automatic machine tool it requires a number of parts and a dedicated mechanism to achieve the purpose so power hacksaw machine parts are as follows the frame or body or base

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