- - (a) mol dm⁻³
 - (c) mol⁻¹ dm³

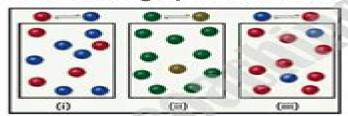
- (b) mol⁻² dm⁶
- (d) √ no unit
- 07. The system is stable in equilibrium when:
 - (a) V Q_c = K_c

(b) Q_c > K_c

(c) Q_c < K

(d) None of these

- 08. Qc can be defined as:
 - (a) ratio of product and reactants
 - (b) V ratio of molar concentration of product and reactant at specific time
 - (c) ratio of molar concentration of product and molar volume of reactant
 - (d) ratio of molar concentration of product and reactant raised to the power of coefficient
- 09. Which of the following represent backward reaction?



(a) (i) and (ii)

(b) (ii) and (iii)

(c) (ii) only

(d) V (iii) only

- 10. The value of K_c increases when:
 - (a) [Product] is less

(b) V [Product] is more

(c) [Reactant] is more

- (d) [Reactant = product]
- 11. Which of these are reversible reactions?
 - (a) Condensation

(b) Evaporation

(c) **V** Condensation & Evaporation

(d) None of them

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