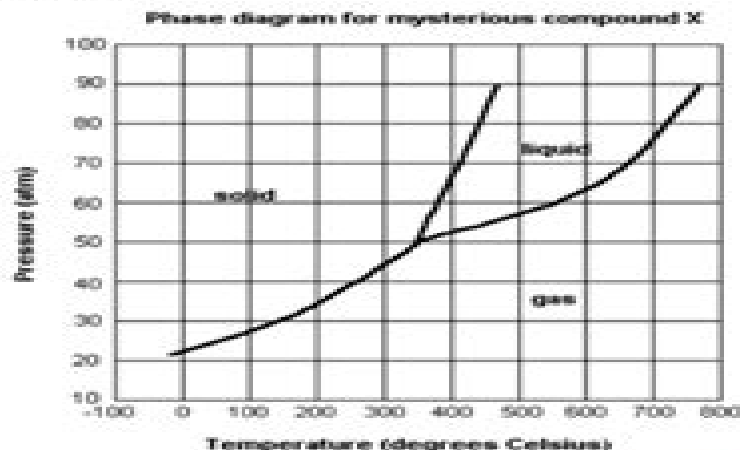


## Phase Diagram Worksheet

For each of the questions on this worksheet, refer to the phase diagram for mysterious compound X.



- 1) What is the critical temperature of compound X?  **$\sim 470^{\circ}\text{C}$**
- 2) If you were to have a bottle containing compound X in your closet, what phase would it most likely be in?  
**Extrapolating from this diagram, it's most likely a gas.**
- 3) At what temperature and pressure will all three phases coexist?  
 **$350^{\circ}\text{C}$ ,  $\sim 51\text{ atm}$**
- 4) If I have a bottle of compound X at a pressure of 45 atm and temperature of  $100^{\circ}\text{C}$ , what will happen if I raise the temperature to  $400^{\circ}\text{C}$ ?  
**It will sublime**
- 5) Why can't compound X be boiled at a temperature of  $200^{\circ}\text{C}$ ?  
**It does not form a liquid at this temperature. It only exists as a liquid at temperatures above  $350^{\circ}\text{C}$ .**
- 6) If I wanted to, could I drink compound X?  
**No. At the temperatures and pressures that it forms a liquid, you'd probably die.**

# Multiple Choice Questions Phase Diagram

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## **Multiple Choice Questions Phase Diagram:**

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**Engineering Chemistry I (for BPUT)** Patrizia Bocchetta, 2024

Introduction to Materials Science and Engineering Yip-Wah Chung, Monica Kapoor, 2022-04-07 Our civilization owes its most significant milestones to our use of materials Metals gave us better agriculture and eventually the industrial revolution



silicon gave us the digital revolution and we're just beginning to see what nanomaterials yield. Updated to reflect the many societal and technological changes in the field since publication of the first edition, *Introduction to Materials Science and Engineering, Second Edition* offers an interdisciplinary view that emphasizes the importance of materials to engineering applications and builds the basis needed to select, modify, and create materials to meet specific criteria. The most outstanding feature of this book is the authors' unique and engaging application-oriented approach. By beginning each chapter with a real-life example, an experiment, or interesting facts, the authors wield an expertly crafted treatment that entertains and motivates as much as it informs and educates. The discipline is linked to modern developments such as semiconductor devices, nanomaterials, and thin films while working systematically from atomic bonding and analytical methods to crystalline, electronic, mechanical, and magnetic properties, as well as ceramics, polymers, corrosion, and phase diagrams. Updates in the Second Edition: References to advances in the field, including computational thermodynamics allowing computation of phase diagrams with great accuracy and new materials; Updated applications and technologies such as electric vehicles and the use of magnetic fields as a processing tool; Revised practical end-of-chapter problems that go beyond traditional plug-and-chug exercises to enhance learning; More examples with detailed solutions in each chapter; A new chapter highlighting how materials can impact four United Nations Sustainable Development Goals. This book is written for undergraduate students and readers interested in introductory materials science and engineering concepts. This concise textbook provides a strong foundation in materials science, engineering, and its applications. A solutions manual and PowerPoint lecture slides are available for adopting professors.

**CliffsNotes AP Chemistry** Bobrow Test Preparation Services, 2009-02-09. The book itself contains chapter-length subject reviews on every subject tested on the AP Chemistry exam, as well as both sample multiple-choice and free-response questions at each chapter's end. Two full-length practice tests with detailed answer explanations are included in the book.

**Cracking the AP Chemistry, 2002-2003 Edition** Paul Foglino, Princeton Review (Firm), 2002-01-29. The Princeton Review realizes that acing the AP Chemistry exam is very different from getting straight A's in school. The Princeton Review doesn't try to teach students everything there is to know about chemistry; only the techniques they'll need to score higher on the exam. There's a big difference. In *Cracking the AP Chemistry*, TPR will teach test takers how to think like the test makers and score higher by reviewing key chemistry concepts. Earn more points by becoming familiar with the format of the test. Safeguard against traps that can lower scores. Perfect skills with review questions in each chapter. This book includes 2 full-length simulated AP Chemistry exams. All of The Princeton Review practice test questions are like the ones test takers will see on the actual exam, and every solution is fully explained.

**Physics** Eugene Hecht, 2003. Student text. An Introduction to Physics. Measurement. The Language of Physics. Kinematics. Speed. Velocity. Speed. Velocity. Relative Motion. Kinematics. Acceleration. The Concept of Acceleration. Uniformly Accelerated Motion. Free Fall. Newton's Three Laws. The Three Laws. Dynamics. Statics. Centripetal Force. Gravity. Centripetal Force. Gravity. The Cosmic Force. Energy.

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