

## 2.12. Answers

1.  $\cdot \dot{\text{C}} \cdot$
- 4 unpaired v.e  $\therefore$  can form 4 covalent bonds.
  - can combine with one or two other  $\text{C}_2 \rightarrow$  straight chains
  - when combined with three or four  $\text{C}_2$ , each secondary C can form a chain  $\therefore$  forms branched chain.
  - due to  $e^-$  repulsion, the bonds form a tetrahedral structure  $\therefore$  the angles b/w the bonds allow chains of C to bend around  $\rightarrow$  ring structures.



2. - to give the characteristic chemical rxns. of molecules
3. their use is dependent on the properties of the compound which are dependent on their structures.
- the same functional groups behave consistently from one organic molecule to another.
4. - funct. groups determines the polarity of a molecule  $\rightarrow$  determines the type of solvents and the type of molecules that are attracted to.
- polar dissolves polar
  - non-polar dissolves non-polar.
5. - Oxygen helps establish the polarity of a molecule
6.  $-\text{COOH} + \text{H}_2\text{O} \rightleftharpoons -\text{COO}^- + \text{H}^+ / \text{H}_3\text{O}^+$
- $-\text{NH}_2 + \text{H}_2\text{O} \rightleftharpoons -\text{NH}_3^+ + \text{OH}^-$   
(H-OH)
- $-\text{H}_2\text{PO}_4 + \text{H}_2\text{O} \rightleftharpoons \text{HPO}_4^- + \text{H}^+ / \text{H}_3\text{O}^+$   
( $\begin{smallmatrix} + \\ \text{H}_2\text{O} \end{smallmatrix} \rightleftharpoons \text{PO}_4^{2-} + \text{H}^+ / \text{H}_3\text{O}^+$ )

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**Sara Freeman**



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francesco borromini ,bɒrə'miːni 1 italian fran'tjesko borro'miːni byname of francesco castelli italian ka'stelli 25 september 1599 2 august 1667 2 was an italian architect born in the modern swiss canton of ticino 3 who with his contemporaries gian lorenzo bernini and pietero da cortona was a leading

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