

KISA Pre Board Examination - Answer key

SUBJECT-CHEMISTRY

Section A

Question 1.

1x15=15M

- i. (a). they have similar electronic configuration in valence shell.
- ii. (c). CuSO_4
- iii. (b). Oxygen is released at anode.
- iv. (c). Substitution
- v. (a). He
- vi. (c). Consists of molecules
- vii. (b). Lead
- viii. (d). Argon
- ix. (a). Magnesium oxide
- x. (b). Aqueous acetic acid
- xi. (c). Reacts to form water, nitrogen dioxide and carbon dioxide
- xii. (a). K_2SO_4
- xiii. (c). 1
- xiv. (c). Fe, Cr and Ni
- xv. (a). Decreases

Question 2.

- (i) (a) A-----→ Conc.sulphuric acid (3)
B-----→ sodium nitrate/potassium nitrate
C-----→ Nitric acid



- (ii) Column A Column B 1x5=5M

- a) Sodium chloride — ionic bond
- b) Ammonium ion — covalent and co-ordinate bond
- c) Electronegativity across the period —I increases
- d) Nonmetallic character down the group — decreases

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