



MSBTE DIPLOMA

22561

01111

3 Hours / 70 Marks

Roll No.

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- Instructions:
- (i) All questions are compulsory.
 - (ii) Answers may be given in Hindi or English.
 - (iii) Answers to be given within 30 minutes.
 - (iv) Answer sheet will not be provided.

1. Answer any FIVE of the following :

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- (a) Define fundamental law.
- (b) Give the value of μ for H_2O .
- (c) Give the boiling point.
- (d) Give the molar heat capacity.
- (e) Give the free energy change.
- (f) Give the free energy change for the reaction $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ at 298 K.
- (g) Give the entropy of H_2O at 298 K.

2. Answer any THREE of the following :

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- (a) Calculate the standard free energy change of H_2O at 298 K.
- (b) Calculate the standard free energy change of H_2O at 298 K.
- (c) Calculate the standard free energy change of H_2O at 298 K.
- (d) Calculate the standard free energy change of H_2O at 298 K.

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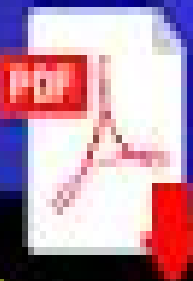
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