PRACTICE PAPER 🔯

SECTION-A (MULTIPLE CHOICE QUESTIONS)

- Read the following statements and choose the correct option.
 - (i) Amines can be distinguished by Hinsberg's reagent.
 - (ii) The three types of amines react similarly with Hinsberg's reagent and hence can be distinguished from each other.
 - (a) (i) is true and (ii) is true.
 - (i) is true and (ii) is false.
 - (c) (i) is false and (ii) is true.
 - (d) (i) is false and (ii) is false.
- 2. One mole of an organic compound A with the formula C,H,O reacts completely with two moles of HI to form X and Y. When Y is boiled with aqueous alkali, it forms Z. Z answers the iodoform test. The compound A is
 - (a) propan-2-ol
- (b) propan-1-ol
- (4d) methoxyethane. (c) ethoxyethane
- 3c 0.30 g of an organic compound containing C. H and O on combustion yields 0.44 g CO, and 0.18 g H.O. If one mol of compound weighs 60, then molecular formula of the compound is
 - (a) C₂H₂O

JOHN C.H.O.

(c) CH₂O

- (d) C₂H₂O
- 1 The correct order of hybridisation of the central atom in the following species NH₃. [PtCl₄]²⁻, PCl₄ and BCl_v is
 - (a) dsp2, dsp2, sp2, sp3, dsp2, dsp2, sp3d, sp2
 - (c) dsp², sp², sp³, dsp³(d) dsp², sp³, sp², dsp³
- Arrange the following compounds in increasing . 9 Consider the following statements : order of oxidation number.

MnCl₂, MnO₂, Mn(OH)₃, KMnO₄

- (a) MnCl₂ < MnO₃ < Mn(OH)₃ < KMnO₄
- (b) MnO₂ < MnCl₂ < KMnO₄ < Mn(OH)₃
- (c) Mn(OH)₁ < MnCl₂ < MnO₂ < KMnO₄
- $MnCl_2 < Mn(OH)_3 < MnO_2 < KMnO_4$

- Elements X, Y and Z have atomic numbers 19. 37 and 55 respectively. Which of the following statements is true about them?
 - (a) Their ionisation potential would increase with increasing atomic number.
 - (b) Y would have an ionisation potential between those of X and Z.
 - (c) Z would have the highest ionisation potential.
 - (d) Y would have the highest ionisation potential.
- Statement-1: Bromobenzene upon reaction with Br₂/Fe gives 1, 4-dibromobenzene as the major product.

Statement-2: In bromobenzene, the inductive effect of the bromo group is more dominant than the mesomeric effect in directing the incoming electrophile.

- (a) Statement-1 is true; statement-2 is true; statement-2. is a correct explanation for statement-1.
- (b) Statement-1 is true; statement-2 is true; statement-2 is not a correct explanation for statement-1.
- Statement-1 is true, statement-2 is false.
 - (d) Statement-1 is false, statement-2 is true.
- What would be the wavelength and name of series respectively for the emission transition for H-atom if it starts from the orbit having radius 1.3225 nm and ends at 211.6 pm?
 - لام) 434 nm, Balmer
- (b) 434 pm. Paschen
- (c) 545 pm, Pfund
- (d) 600 nm, Lyman
- - CH₃OCH₂ is more stable than CH₃CH₂.
 - (II) Me₃CH is more stable than CH₃CH₃CH₃.
 - (III) $CH_1 = CH CH_2$, is more stable than CH₃CH₃CH₃.
 - (IV) CH2 = CH is more stable than CH2CH2.

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