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Johnson

OUTBOARD MOTOR

SERVICE INSTRUCTION BOOK



6 HP



OUTBOARD MARINE BELGIUM, S.A. BRUGES, BELGIUM

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 $2 \text{NaCl} + \text{F}_2 \rightarrow 2 \text{NaF} + \text{Cl}_2$
 $2 \text{H}_2 + \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$
 $\text{Pb} + \text{OH}^- \rightarrow \text{HCl} + \text{H}_2\text{O}$
 $\text{PbCl}_2 + \text{AlBr}_3 + 3 \text{K}_2\text{SO}_4 \rightarrow 6 \text{KBr} + \text{Al}_2\text{SO}_4 + 3 \text{CH}_4 + 2 \text{O}_2$
 $\text{CO}_2 + 2 \text{H}_2\text{O} \rightarrow \text{C}_3\text{H}_8 + 5 \text{O}_2$
 $3 \text{CO}_2 + 4 \text{H}_2\text{O} \rightarrow \text{C}_8\text{H}_{18} + 25 \text{O}_2$
 $16 \text{CO}_2 + 18 \text{H}_2\text{O}$

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 $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF} + \text{Cl}_2$
 $4\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 $5\text{Pb} + 2\text{HCl} \rightarrow 2\text{H}_2 + \text{PbCl}_2$
 $2\text{Al} + 3\text{K}_2\text{SO}_4 \rightarrow 4\text{KBr} + \text{Al}_2\text{SO}_4 + 3\text{H}_2$
 $4\text{CH}_4 + \text{O}_2 \rightarrow 2\text{CO}_2 + 8\text{H}_2\text{O}$
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 $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF} + \text{Cl}_2$
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 $4\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 $5\text{Pb} + 2\text{HCl} \rightarrow 2\text{H}_2 + \text{PbCl}_2$
 $2\text{Al} + 3\text{K}_2\text{SO}_4 \rightarrow 4\text{KBr} + \text{Al}_2\text{SO}_4 + 3\text{H}_2$
 $4\text{CH}_4 + \text{O}_2 \rightarrow 2\text{CO}_2 + 8\text{H}_2\text{O}$
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 $2\text{Fe} + 3\text{H}_2\text{SO}_4 \rightarrow \text{Fe}_2(\text{SO}_4)_3 + 3\text{H}_2$
 $2\text{C}_2\text{H}_6 + 7\text{O}_2 \rightarrow 4\text{CO}_2 + 6\text{H}_2\text{O}$

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 $3\text{Fe} + 2\text{O}_2 \rightarrow \text{Fe}_3\text{O}_4$
 $2\text{Fe} + 3\text{CO} \rightarrow \text{Fe}_2\text{O}_3 + 3\text{C}$
 $4\text{Ba} + 3\text{O}_2 \rightarrow 2\text{Ba}_2\text{O}_3$
 $2\text{Mg} + \text{N}_2 \rightarrow 2\text{Mg}_3\text{N}_2$
 $4\text{Mg} + 7\text{N}_2 \rightarrow 2\text{Mg}_3\text{N}_2 + 4\text{N}_2$
 $2\text{Al} + 3\text{H}_2\text{SO}_4 \rightarrow \text{Al}_2(\text{SO}_4)_3 + 3\text{H}_2$
 $4\text{Al} + 3\text{H}_2\text{SO}_4 \rightarrow 2\text{Al}_2\text{O}_3 + 3\text{H}_2$
 $10\text{HCl} + \text{PbO}_2 \rightarrow 2\text{H}_2\text{O} + \text{PbCl}_2 + 4\text{H}_2\text{O}$

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