

Chemistry – Module 1 – The Chemical Earth

1. The living and nonliving components of Earth contain mixtures

- **construct word and balanced equations of chemical reactions as they are encountered**

General Reactions

a) Single displacement reaction: $AB + C \rightarrow AC + B$



b) Double displacement reaction: $AB + CD \rightarrow AD + CB$



c) Combination reaction: $A + B \rightarrow AB$



d) Decomposition reaction: $AB \rightarrow A + B$

- Heating
- Visible light or UV
- Electrolysis

Common Reactions

a) Metal + Acid \rightarrow Salt + Hydrogen gas

b) Metal + Water \rightarrow Metal hydroxide + water

c) Metal + Salt \rightarrow New Salt + Metal

d) Acid + Base \rightarrow Salt + Water

e) Acid + Carbonate \rightarrow Salt + Carbon dioxide + water

f) Salt + salt \rightarrow New salts

g) Metal Oxide + water \rightarrow Acid

h) Combustion

i. Element + Oxygen \rightarrow Element Oxide

ii. Hydrocarbon + Oxygen \rightarrow Carbon dioxide + Water

i) Heat + Carbonate \rightarrow Metal Oxide + Carbon Dioxide

- **Identify the difference between elements, compounds and mixtures in terms of the particle theory**

All matter is made of continuously moving particles.

An element is a pure substance made up of only one type of atom. It cannot be broken down into smaller substances using ordinary physical and chemical means. All elements are homogeneous (means they have uniform composition throughout). Some elements include aluminium, copper, gold etc.

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SJ Ball



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