ework for Unit #1: Chapter 2

em 2.65

Bicarbonate ($HCO_3^-$), $K_a = 6.3 \times 10^{-11}$

**Express your answers as integers separated by a comma.**

between

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**Part D**

Hydrogen cyanide ($HCN$), $K_a = 7.9 \times 10^{-10}$

**Express your answers as integers separated by a comma.**

between

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**Part E**

Formic acid ($HCOOH$), $K_a = 2.0 \times 10^{-4}$

**Express your answers as integers separated by a comma.**

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Jacquelyn G. Black,George Allan Wistreich,Black

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