

## CHEMICAL THERMODYNAMICS

- 1) In open system, system and surroundings exchange.....  
 (a) Energy only (b) Matter only  
**(c) Energy and matter** (d) Neither energy nor matter
- 2) An isothermal expansion takes place at.....  
 (a) constant pressure (b) constant volume  
**(c) constant temperature** (d) constant temperature & pressure
- 3) In a closed thermos flask few cubes of ice are placed. It is example of.....  
 (a) closed system (b) open system  
**(c) isolated system** (d) thermodynamics system
- 4) In an adiabatic compression, the internal energy of the system.....  
 (a) decreases **(b) increases** (c) increases or decreases (d) none
- 5) Identify the intensive quantities from the following.....  
 (a) volume (b) enthalpy **(c) pressure** (d) none
- 6) Which of the following is extensive property?  
 (a) mole (b) mass (c) volume **(d) all are correct**
- 7) The heat change in a chemical reaction at constant pressure is.....  
**(a)  $\Delta H$**  (b)  $\Delta U$  (c)  $\Delta T$  (d)  $\Delta V$
- 8)  $\Delta U$  is always positive when.....  
**(a) system absorbs heat and work is done on it**  
 (b) system emits heat and work is done on it  
 (c) system emits heat and no work is done on it  
 (d) system absorbs heat and work is done by it
- 9) If in a chemical reaction, the products have less energy than the reactants, that reaction is.....  
 (a) a reversible reaction **(b) an exothermic reaction**  
 (c) an isothermal reaction (d) an endothermic reaction
- 10) A small piece of Na metal is dropped into water. Na bursts into flame, the reaction is:  
 (a) neutralisation (b) reversible **(c) exothermic** (d) endothermic
- 11) For the reaction:  $C_3H_8(g) + 5O_2(g) \rightarrow 3CO_2(g) + 4H_2O(l)$  at constant temperature,  $\Delta H - \Delta U$  is  
 (a)  $+RT$  **(b)  $-3RT$**  (c)  $+3RT$  (d)  $-RT$
- 12) When  $NH_4Cl$  is dissolved in water, the solution becomes cold. The change is?  
**(a) endothermic** (b) exothermic (c) super cooling (d) none

# Mcqs Of Thermodynamics With Answers

**R P Meena**



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Friction Drag and Centripetal Force MCQ Chapter 14 Fundamental Constants of Physics MCQ Chapter 15 Geometric Optics MCQ Chapter 16 Inductance MCQ Chapter 17 Kinetic Energy MCQ Chapter 18 Longitudinal Waves MCQ Chapter 19 Magnetic Force MCQ Chapter 20 Models of Magnetism MCQ Chapter 21 Newton s Law of Motion MCQ Chapter 22 Newtonian Gravitation MCQ Chapter 23 Ohm s Law MCQ Chapter 24 Optical Diffraction MCQ Chapter 25 Optical Interference MCQ Chapter 26 Physics and Measurement MCQ Chapter 27 Properties of Common Elements MCQ Chapter 28 Rotational Motion MCQ Chapter 29 Second Law of Thermodynamics MCQ Chapter 30 Simple Harmonic Motion MCQ Chapter 31 Special Relativity MCQ Chapter 32 Straight Line Motion MCQ Chapter 33 Transverse Waves MCQ Chapter 34 Two and Three Dimensional Motion MCQ Chapter 35 Vector Quantities MCQ Chapter 36 Work Kinetic Energy Theorem MCQ The Alternating Fields and Currents MCQ PDF e Book Chapter 1 practice test to solve MCQ questions on Alternating current damped oscillations in an RLS circuit electrical mechanical analog forced and free oscillations LC oscillations phase relations for alternating currents and voltages power in alternating current circuits transformers The Astronomical Data MCQ PDF e Book Chapter 2 practice test to solve MCQ questions on Aphelion distance from earth eccentricity of orbit equatorial diameter of planets escape velocity of planets gravitational acceleration of planets inclination of orbit to earth s orbit inclination of planet axis to orbit mean distance from sun to planets moons of planets orbital speed of planets perihelion period of rotation of planets planet densities planets masses sun earth and moon The Capacitors and Capacitance MCQ PDF e Book Chapter 3 practice test to solve MCQ questions on Capacitor in parallel and in series capacitor with dielectric charging a capacitor cylindrical capacitor parallel plate capacitor The Circuit Theory MCQ PDF e Book Chapter 4 practice test to solve MCQ questions on Loop and junction rule power series and parallel resistances single loop circuits work energy and EMF The Conservation of Energy MCQ PDF e Book Chapter 5 practice test to solve MCQ questions on Center of mass and momentum collision and impulse collisions in one dimension conservation of linear momentum conservation of mechanical energy linear momentum and Newton s second law momentum and kinetic energy in collisions Newton s second law for a system of particles path independence of conservative forces work and potential energy The Coulomb s Law MCQ PDF e Book Chapter 6 practice test to solve MCQ questions on Charge is conserved charge is quantized conductors and insulators and electric charge The Current Produced Magnetic Field MCQ PDF e Book Chapter 7 practice test to solve MCQ questions on Ampere s law and law of Biot Savart The Electric Potential Energy MCQ PDF e Book Chapter 8 practice test to solve MCQ questions on Introduction to electric potential energy electric potential and equipotential surfaces The Equilibrium Indeterminate Structures MCQ PDF e Book Chapter 9 practice test to solve MCQ questions on Center of gravity density of selected materials of engineering interest elasticity equilibrium indeterminate structures ultimate and yield strength of selected materials of engineering interest and Young s modulus of selected materials of engineering interest The Finding Electric Field MCQ PDF e Book Chapter 10 practice test to solve MCQ questions on Electric field electric field due to

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*Comprehensive Objective Physics* Narinder Kumar, Dr. J. K. Juneja, 2006

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*Problems In Chemical Thermodynamics, With Solutions* Maka Aleksishvili, Shota Sidamonidze, 2002-11-26 The methods of chemical thermodynamics are effectively used in many fields of science and technology Mastering these methods and their use in practice requires profound comprehension of the theoretical questions and acquisition of certain calculating skills This book is useful to undergraduate and graduate students in chemistry as well as chemical thermal and refrigerating technology it will also benefit specialists in all other fields who are interested in using these powerful methods in their practical activities

*Basic Thermodynamics* B. K. VENKANNA, SWATI B. V., B. K. Venkanna, B. V. Swati, 2010-07 This book provides an in depth discussion of the principles of thermodynamics It focuses on engineering applications of theory and sound techniques for solving thermodynamic problems The book presents the fundamental concepts of thermodynamics and describes the theory of work and heat The text covers in detail the first law and the second law of thermodynamics with their applications It also explains the concepts of entropy and availability and irreversibility In addition the book presents thermodynamic properties of pure substances ideal gases and mixtures of ideal gases as well as real gases This book is designed for undergraduate students of mechanical engineering industrial and production engineering automobile engineering and aeronautical engineering for their courses in thermodynamics Key Features Presents the text in a simple and elegant manner to enable the students to grasp the essentials of the subject easily and quickly Covers all types of problems of various difficulty levels Includes more than 300 worked out examples and a large number of end of chapter exercises Provides solutions to several

model question papers at the end of the book      **Thermodynamics** Rumi Michael Leigh, 2023-05-24 Thermodynamics Things You Should Know Questions and Answers explains the principles of thermodynamics in a question and answer format The book introduces energy work heat and the basic laws of thermodynamics It describes the concepts of entropy enthalpy internal energy and equilibrium and shows how these ideas connect to physical and chemical processes Key topics include the first and second laws of thermodynamics the role of entropy in natural systems and the behavior of gases as described by equations of state The text also examines heat transfer efficiency and applications of thermodynamics in engines refrigeration and power systems Worked examples and explanations show how thermodynamic concepts are applied in solving problems and understanding physical phenomena This book will interest students engineering learners and those studying physics who want to understand thermodynamics and its applications      Problems And Solutions On Thermodynamics And Statistical Mechanics Yung-kuo Lim, 1990-02-01 The material for these volumes has been selected from the past twenty years examination questions for graduate students at University of California at Berkeley Columbia University the University of Chicago MIT State University of New York at Buffalo Princeton University and University of Wisconsin

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## **Table of Contents Mcqs Of Thermodynamics With Answers**

1. Understanding the eBook Mcqs Of Thermodynamics With Answers
  - The Rise of Digital Reading Mcqs Of Thermodynamics With Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Mcqs Of Thermodynamics With Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Mcqs Of Thermodynamics With Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Mcqs Of Thermodynamics With Answers
  - Personalized Recommendations

- Mcqs Of Thermodynamics With Answers User Reviews and Ratings
- Mcqs Of Thermodynamics With Answers and Bestseller Lists
- 5. Accessing Mcqs Of Thermodynamics With Answers Free and Paid eBooks
  - Mcqs Of Thermodynamics With Answers Public Domain eBooks
  - Mcqs Of Thermodynamics With Answers eBook Subscription Services
  - Mcqs Of Thermodynamics With Answers Budget-Friendly Options
- 6. Navigating Mcqs Of Thermodynamics With Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Mcqs Of Thermodynamics With Answers Compatibility with Devices
  - Mcqs Of Thermodynamics With Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Mcqs Of Thermodynamics With Answers
  - Highlighting and Note-Taking Mcqs Of Thermodynamics With Answers
  - Interactive Elements Mcqs Of Thermodynamics With Answers
- 8. Staying Engaged with Mcqs Of Thermodynamics With Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Mcqs Of Thermodynamics With Answers
- 9. Balancing eBooks and Physical Books Mcqs Of Thermodynamics With Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Mcqs Of Thermodynamics With Answers
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Mcqs Of Thermodynamics With Answers
  - Setting Reading Goals Mcqs Of Thermodynamics With Answers
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Mcqs Of Thermodynamics With Answers
  - Fact-Checking eBook Content of Mcqs Of Thermodynamics With Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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