

# June 2010 Chemistry Regents

## Exam Part 2 – with answers

- 51 What is the total number of electron pairs shared between the carbon atom and one of the oxygen atoms in a carbon dioxide molecule?
- 52 Explain, in terms of subatomic particles, why the radius of a chloride ion is larger than the radius of a chlorine atom.
- 53 Explain, in terms of valence electrons, why the bonding in magnesium oxide,  $\text{MgO}$ , is similar to the bonding in barium chloride,  $\text{BaCl}_2$ .

# June 2012 Chemistry Regents Answers

**Sebastian Brünink**



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