

## ACIDS AND BASES QUIZ

1. Which of the following indicates an acidic solution?

- A) pH = 7.5                      B) pH = 7                      C) pH = 6.5  
D) more than one correct response      E) no correct response

2. Which of the following equations represents an acid-base neutralization reaction?

- A)  $\text{H}_2\text{SO}_4 + \text{Zn} \rightarrow \text{ZnSO}_4 + \text{H}_2$                       B)  $\text{HNO}_3 + \text{KOH} \rightarrow \text{KNO}_3 + \text{H}_2\text{O}$   
C)  $\text{Ba}(\text{OH})_2 + \text{Na}_2\text{SO}_4 \rightarrow \text{BaSO}_4 + 2\text{NaOH}$       D) more than one correct response  
E) none of the above

3. In which of the following pairs of substances are both members of the pair salts?

- A) NaCl and NaOH                      B)  $\text{KNO}_3$  and  $\text{K}_2\text{SO}_4$                       C)  $\text{H}_3\text{PO}_4$  and  $(\text{NH}_4)_3\text{PO}_4$   
D) more than one correct response                      E) none of the above

4. If the pH of a solution increases from 4.0 to 6.0, the hydronium ion concentration

- A) increases by a factor of 1.5.                      B) increases by a factor of 100.  
C) decreases by a factor of 1.5.                      D) decreases by a factor of 100.

5. Which of the following statements concerning Arrhenius acids and Arrhenius bases is incorrect?

- A) In the pure state, Arrhenius acids are covalent compounds.  
B) In the pure state, Arrhenius bases are ionic compounds.  
C) Dissociation is the process by which Arrhenius acids produce  $\text{H}^+$  ions in solution.  
D) Arrhenius bases are also called oxides.

6. Highly basic solutions are:

- A) safe to handle                      B) sweet tasting  
C) corrosive and reactive like highly acidic solutions                      D) corrosive but not reactive

7. When dissolved in water, sodium hydroxide (NaOH):

- A) turns phenolphthalein colourless.      B) produces a high concentration of hydronium ions.  
C) releases  $\text{O}^{2-}$  ions.                      D) produces a high concentration of hydroxide ions.  
E) reacts with carbonates to produce a salt and carbon dioxide gas.

8. Bases taste:

- A) Salty                      B) Bitter                      C) Sweet                      D) Sour

9. The characteristic ion of acids is:

- A) carbonate ion  $\text{CO}_3^{2-}$       B) phosphate ion  $\text{PO}_4^{3-}$       C) hydrogen ion ( $\text{H}^+$ )      D) hydroxide ion ( $\text{OH}^-$ )

10. Which of the following pH values indicates the strongest base?

- A) 4                      B) 7                      C) 11                      D) 1

11. A positive test for a base occurs when:

# Multiple Choice On Acids And Bases General Chem

**J Ma**



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**General Chemistry** James E. Brady, Gerard E. Humiston, 1982 The Fifth Edition retains the pedagogical strengths that made the previous editions so popular and has been updated reorganized and streamlined Changes include more accessible introductory chapters with greater stress on the logic of the periodic table earlier introduction of redox reactions greater emphasis on the concept of energy a new section on Lewis structures earlier introduction of the ideal gas law and a new development of thermodynamics Each chapter ends with review questions and problems

**Fundamentals of General, Organic, and Biological Chemistry** Daniel Sullivan (Prof. of Chemistry.), 1999

**MSCEIS 2019** Lala Septem Riza, Eka Cahya Prima, Toni Hadibarata, Peter John Aubusson, 2020-07-30 The 7th Mathematics Science and Computer Science Education International Seminar MSCEIS was held by the Faculty of Mathematics and Natural Science Education Universitas Pendidikan Indonesia UPI and the collaboration with 12 University associated in Asosiasi MIPA LPTK Indonesia AMLI consisting of Universitas Negeri Semarang UNNES Universitas Pendidikan Indonesia UPI Universitas Negeri Yogyakarta UNY Universitas Negeri Malang UM Universitas Negeri Jakarta UNJ Universitas Negeri Medan UNIMED Universitas Negeri Padang UNP Universitas Negeri Manado UNIMA Universitas Negeri Makassar UNM Universitas Pendidikan Ganesha UNDHAKSA Universitas Negeri Gorontalo UNG and Universitas Negeri Surabaya UNESA In this year MSCEIS 2019 takes the following theme Mathematics Science and Computer Science Education for Addressing Challenges and Implementations of Revolution Industry 4.0 held on October 12 2019 in Bandung West Java Indonesia

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**The Success Manual for General Chemistry** Elizabeth Kean, Catherine Middlecamp, 1986

**Study Guide, Chemical Principles, Fifth Edition, Zumdahl** Paul B. Kelter, Steven S. Zumdahl, 2004-04

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Saunders Interactive General Chemistry CD-ROM Mary L Kotz, John C. Kotz, William J. Vining, Justin T. Fermann, 1998-11

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